

Impact of mineral and non-mineral sources of iron and sulfur on the metalloproteome of *Methanosarcina barkeri*

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ABSTRACT Methanogens often inhabit sulfidic environments that favor the precipitation of transition metals such as iron (Fe) as metal sulfides, including mackinawite (FeS) and pyrite (FeS₂). These metal sulfides have historically been considered biologically unavailable. Nonetheless, methanogens are commonly cultivated with sulfide (HS⁻) as a sulfur source, a condition that would be expected to favor metal precipitation and thus limit metal availability. Recent studies have shown that methanogens can access Fe and sulfur (S) from FeS and FeS₂ to sustain growth. As such, medium supplied with FeS₂ should lead to higher availability of transition metals when compared to medium supplied with HS⁻. Here, we examined how transition metal availability under sulfidic (i.e., cells provided with HS⁻ as sole S source) versus non-sulfidic (cells provided with FeS₂ as sole S source) conditions impact the metalloproteome of *Methanosarcina barkeri* Fusaro. To achieve this, we employed size exclusion chromatography coupled with inductively coupled plasma mass spectrometry and shotgun proteomics. Significant changes were observed in the composition and abundance of iron, cobalt, nickel, zinc, and molybdenum proteins. Among the differences were alterations in the stoichiometry and abundance of multisubunit protein complexes involved in methanogenesis and electron transport chains. Our data suggest that *M. barkeri* utilizes the minimal iron-sulfur cluster complex and canonical cysteine biosynthesis proteins when grown on FeS₂ but uses the canonical Suf pathway in conjunction with the tRNA-Sep cysteine pathway for iron-sulfur cluster and cysteine biosynthesis under sulfidic growth conditions.

IMPORTANCE Proteins that catalyze biochemical reactions often require transition metals that can have a high affinity for sulfur, another required element for life. Thus, the availability of metals and sulfur are intertwined and can have large impacts on an organism's biochemistry. Methanogens often occupy anoxic, sulfide-rich (euxinic) environments that favor the precipitation of transition metals as metal sulfides, thereby creating presumed metal limitation. Recently, several methanogens have been shown to acquire iron and sulfur from pyrite, an abundant iron-sulfide mineral that was traditionally considered to be unavailable to biology. The work presented here provides new insights into the distribution of metalloproteins, and metal uptake of *Methanosarcina barkeri* Fusaro grown under euxinic or pyritic growth conditions. Thorough characterizations of this methanogen under different metal and sulfur conditions increase our understanding of the influence of metal availability on methanogens, and presumably other anaerobes, that inhabit euxinic environments.

KEYWORDS iron acquisition, archaea, metallomics, metal regulation, proteomics, metalloproteins, methanogenesis, methanogens

Methanogens conserve energy through the metabolism of hydrogen (H₂), organic acids, or small methylated compounds, generating methane (CH₄) as a byproduct

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(1). Metabolism of these compounds is achieved through a conserved central methanogenesis pathway, with various add-on enzymes that enable the metabolism of diverse substrates (1). Methanogen metabolism is heavily reliant on metals, with methanogens requiring significantly more iron (Fe) per cell than other organisms based upon increased abundance of iron-sulfur ([Fe-S]) cluster motifs encoded in their genome (2). Beyond Fe, methanogens also require relatively large amounts of nickel (Ni), cobalt (Co), zinc (Zn), and molybdenum (Mo) compared to other prokaryotic organisms (3). Cobalt is a necessary metal cofactor in corrinoid methyltransferase proteins utilized in the methylotrophic pathway (4, 5). Previous work has shown that methanogenic activity in sludge bed reactors increased with the addition of Co when methanol was provided as substrate (6). Nickel is largely used in various [Ni-Fe] hydrogenases for the activation of H₂, and is additionally present in the tetrapyrrole cofactor F₄₃₀ in methyl-coenzyme M reductase (MCR) (7, 8). Zinc is found in the ribosomal complexes of methanogens (3, 9), whereas Mo is used in enzymes to catalyze numerous redox reactions, including molybdenum nitrogenase and formylmethanofuran dehydrogenase (10–13). The prevalence of metals in methanogen metabolism motivates research to understand how metal bioavailability impacts their physiology and thus ecology.

Methanogens are common in environments that lack oxygen but are rich in sulfide (HS⁻), also known as euxinic conditions (14). Euxinic conditions promote sulfidation of transition metals, ultimately resulting in the formation of sulfide minerals such as mackinawite (FeS) and pyrite (FeS₂) (15–18). Recently, representatives of both type 1 (i.e., *Methanococcus voltae* A3, *Methanococcus maripaludis* S2) and type 2 (*Methanosarcina barkeri* MS and Fusaro) methanogens have been shown to be capable of meeting their Fe and sulfur (S) requirements through the reductive dissolution of FeS₂ (14, 19, 20). However, the exact mechanism for Fe and S acquisition has not been fully elucidated. Comparative proteomics of type 1 (*M. voltae* A3) and type 2 (*M. barkeri* MS) methanogens grown on soluble Fe(II)/HS⁻ or FeS₂ as the sole Fe and S sources, respectively, show major changes in the proteomes of these organisms, in particular proteins involved in metal binding, metal uptake, and predicted Fe-trafficking and Fe-storage pathways (21, 22). Appendages attaching *M. voltae* cells to FeS₂ have been observed with field emission scanning electron microscopy when grown on FeS₂ mineral (20). Additional experiments with *M. barkeri* Fusaro demonstrated that the methanogen was unable to grow when cells were prevented physical access to FeS₂ (19). However, growth was supported when an exogenous electron shuttle (a synthetic quinone analog) was added to the growth medium (19). These data support a hypothesis wherein cells must either first attach to the mineral surface to facilitate extracellular electron transfer to facilitate mineral reduction or secrete electroactive shuttle compounds to facilitate mineral reduction (20).

Reduction of FeS₂ releases HS⁻ into solution and forms pyrrhotite (Fe_{1-x}S) on the mineral surface (19). Fe(II) (but not HS⁻) then dissolves from Fe_{1-x}S and reacts with the previously released HS⁻ to form aqueous FeS clusters (FeS_{aq}) that are predicted to be neutrally charged and thought to passively diffuse or be actively transported (if charged) across the membrane. After assimilation of FeS_{aq} into cells, Fe and S are trafficked and stored by largely uncharacterized mechanisms that may involve the formation of intracellular thioferrate-like minerals (14). Nonetheless, cells can synthesize protein-bound [Fe-S] clusters when grown on FeS₂ (14). Biosynthesis of [Fe-S] clusters in methanogens generally occurs via the sulfur (SUF) system in type I (non-cytochrome-encoding) methanogens or the SUF system and iron-sulfur cluster assembly (ISC) pathway in type II (cytochrome-encoding) methanogens (23, 24). It has previously been shown that all methanogens have the genetic capacity to generate [Fe-S] clusters through a minimal SUF system (SufCB). This system is comprised of an ATPase, SufC, and a scaffold protein, SufB (23). A recent phylogenetic analysis of over 10,000 archaeal and bacterial genomes revealed an additional minimalistic pathway called the minimal [Fe-S] cluster assembly (MIS) which requires fewer genes than the ISC system to synthesize and traffic [Fe-S] clusters with ISC-like proteins using a cysteine desulfurase, Mis-S, and a scaffold protein, Mis-U (25). Type II methanogens can encode both minimal [Fe-S]

cluster synthesis pathways, posing new questions with respect to what types of growth conditions support the use of each of these [Fe-S] cluster biosynthetic pathways.

To begin to probe this question, and to better understand how growth with FeS₂ versus Fe(II)/HS⁻ influences the availability of transition metals and their use in the metalloproteome of methanogens, we employed size exclusion chromatography (SEC) in line with an inductively coupled plasma mass spectrometer (ICPMS) to detect changes in metalloproteins in cultures of *M. barkeri* Fusaro. In parallel, size exclusion chromatography was also used for fractionation of the proteome and the fractions were subsequently analyzed by tandem mass spectrometry for protein identification. This work shows that changes in the metalloproteome of *M. barkeri* cultured under the sulfidic conditions typically used to culture methanogens and those cultured with FeS₂ are surprisingly large for several transition metals, and include apparent major changes in the mechanism by which cells synthesize [Fe-S] clusters, synthesize cysteine, and generate a membrane potential.

MATERIALS AND METHODS

Culturing *Methanosarcina barkeri* Fusaro

Methanosarcina barkeri strain Fusaro (ATCC-BAA-2329) was grown in 75 mL of base salts medium as previously described (19). Acetate (40 mM) and methanol (0.5% vol/vol) were used as methanogenesis substrates under a headspace of N₂:CO₂ (80%:20%) pressurized to 2.5 bar. Growth medium was prepared without Fe and S in MilliQ water and using acid (10% HNO₃) washed glassware. Medium was amended with sulfide, added as Na₂S, to a final concentration of 1 mM and ferrous iron [Fe(II) added as FeCl₂] to a final concentration of 20 μM. Pyrite was synthesized as previously described (20). FeS₂ was added as a slurry to a final concentration of 2 mM Fe. Cultures were incubated in serum bottles statically on their sides and in the dark at 38°C. Cultures were inoculated from an exponentially growing culture. Once they reached early log phase, the cells were harvested under anoxic headspace by centrifugation at 4,600 × *g* for 30 min at 4°C. Then, the supernatant was decanted and cell pellet was flash frozen in liquid N₂. Frozen cell pellets were stored at -80°C until further analysis. Two control experiments were performed in the same fashion. The first control experiment was an abiotic experiment where growth medium and FeS₂ were incubated together. The second control experiment consisted of *M. barkeri* grown on FeS₂ (as above) but with exposure to 2 mM Na₂S for 15 min immediately prior to harvesting.

Methane (CH₄) in the headspace was quantified via an SRI 8610C gas chromatograph (SRI Instrument Company Inc., Torrance, CA) with the instrument injector valve set at 55°C using a Hayesep DB 100/120 packed column (Valco Instrument Company Inc., Houston, TX) with the oven set at 44°C. Helium (He) was used as the carrier gas and CH₄ was detected via a flame ionization detector set at 156°C. Standard curves to relate CH₄ peak area to concentration were made each day using certified standards (EGAS Depot, Nampa, ID). The concentration of HS⁻ (product of FeS₂ reduction) was determined colorimetrically on subsamples of culture via the methylene blue assay using a Genesys 10S Vis Spectrophotometer (ThermoScientific, Waltham, MA) (26).

M. barkeri grows in aggregates and this can confound direct enumeration of cells via microscopy (27). As such, growth of *M. barkeri* was monitored by quantification of genomic DNA as a proxy for cell production. To quantify DNA, 1 mL subsamples of *M. barkeri* Fusaro cultures were collected via sterile syringe and needle, transferred to sterile microfuge tubes, centrifuged at 20,000 × *g* for 20 min at 4°C, and then the supernatant was decanted. After centrifugation, the cell pellet was resuspended in a lysis buffer solution containing 489 μL sodium phosphate buffer (MP Biomedicals, Irvine, CA, USA) and 61 μL MT buffer (MP Biomedicals). The mixture was then gently stirred before three cycles of freezing at -80°C and heating/thawing at 52°C in a heat block. The mixture was then transferred to a 2 mL screw cap bead beating tube (MP Biomedicals) and homogenized for 40 s using a bead beater (Biospec Products, Bartlesville, OK, USA). To

separate DNA from cell debris, bead beating tubes were centrifuged for 15 min at $14,000 \times g$ at 4°C . The concentration of DNA was quantified using a Qubit HS Double-Stranded DNA kit and Qubit fluorimeter (Invitrogen, Carlsbad, CA, USA).

Protein extraction

Cell pellets were washed three times by gentle manual shaking in 5 mL of lysis solution (N_2 sparged 200 mM ammonium acetate; pH 8). Following each wash, cells were pelleted by centrifugation at $1,500 \times g$ at 4°C for 5 min. Cell pellets were then resuspended in 800 μL of lysis buffer and lysed with matrix E (MPBio) on a FastPrep-24 5G bead beater (MPBio) using four 30-s cycles at a speed of 5.5 m/s with 1-min rest at room temperature ($\sim 21^{\circ}\text{C}$) in between cycles. Lysates were pelleted at $18,000 \times g$ at 4°C for 20 min and supernatants were then filtered over 0.2 μm cellulose acetate filters using spin-X centrifuge tubes (Costar) at $24,700 \times g$. Protein concentrations were determined using the Bradford assay. Samples were analyzed immediately after cell lysis to minimize exposure to air.

Metalloproteome analysis

The metalloproteomes of *M. barkeri* Fusaro cells grown with FeS_2 or Fe(II)/HS^- were analyzed via SEC-ICPMS using an Agilent Infinity II LC coupled to an Agilent 7800 ICPMS, as previously described (28). Briefly, 95 μg of soluble protein was injected onto an Agilent Bio-SEC 3 (3 μm , 300 \AA , 4.6×300 mm) column and proteins were separated for 25 min in N_2 sparged 200 mM ammonium acetate (pH 8) as the mobile phase with a flow rate of 0.4 mL/min. Ammonium acetate was used as the mobile phase because of its compatibility with SEC and ICPMS and its ability to not disrupt non-covalent interactions (29). The signals of ^{56}Fe , ^{59}Co , ^{60}Ni , ^{66}Zn , and ^{95}Mo were collected with an integration time of 1.5 s per element. Monitoring of ion signals was performed in Agilent MassHunter 4.7 (version C.01.06). The ICPMS parameters were as follows: RF power 1550 W, RF Matching 1.00 V, nebulizer Gas 0.99 L/min, option gas 0.0%, nebulizer pump 0.30 rps, S/C temp 2°C , makeup gas 0.00 L/min, extract 1 lens 0.0 V, extract 2 lens -195.0 V, omega bias -95 V, omega lens 8.4 V, cell entrance -40 V, cell exit -60 V, deflect 0 V, plate bias -55 V, He flow rate 4.0 mL/min, octupole bias -18 V, octupole RF 200 V, and energy discrimination 3.0 V. SEC-ICPMS data were normalized to the highest peak intensity for each metal.

ICPMS analysis of soluble cellular metals

To assess total cellular soluble metal content, 178 μg of protein was digested in 8% (wt/vol) optima grade HNO_3 (Fisher Chemical) for 20 min at 99°C . After digestion, samples were centrifuged at $16,000 \times g$ at 4°C for 5 min and the supernatant was collected. Quantitative analysis was performed using an Agilent Infinity II autosampler coupled to an Agilent 7800 ICPMS in helium mode. One hundred microliters of supernatant was infused with the mobile phase (2% HNO_3 , 0.5% HCl in water) with a flow rate of 1 mL/min. The signals of five metal ions were monitored for 30 s and included ^{56}Fe , ^{59}Co , ^{60}Ni , ^{66}Zn , and ^{95}Mo with an integration time/mass of 0.5 s per analyte.

Abiotic trace metal availability

An abiotic experiment was conducted to investigate the effect of different culture conditions on the availability of transition metals. The experiment was conducted in basal medium that lacked vitamins and methanol/acetate. Three different conditions were tested: soluble Fe(II) (20 μM) alone, soluble Fe(II) (20 μM) and sulfide (1 mM), or FeS_2 (2 mM). The trace element stock solution contained (g L^{-1}) ZnCl_2 , 0.070; $\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$, 0.100; H_3BO_3 , 0.006; $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$, 0.190; $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$, 0.002; $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$, 0.024; $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$, 0.036. All three experiments were amended with trace elements to a final concentration of 1 mL L^{-1} . After the amendments were added, the reactors were over-pressurized with 15 psi of 80:20 $\text{N}_2:\text{CO}_2$. The reactors were then sampled via syringe and needle (~ 1 hour after stock additions) and 3 mL of solution was filtered with a

0.22 μm polytetrafluoroethylene (PTFE) membrane (ThermoFisher, Titan filters) into clean 15 mL tubes which were then frozen (t_0). The reactors were then incubated at 37°C for 4 days, after which the reactors were again sampled and the filtrates frozen (t_f). The filtrates were thawed and then acidified with trace-metal grade HNO_3 to 3% vol/vol. The samples were diluted directly prior to trace metal quantification via ICPMS to prevent salt-induced signal suppression.

SEC fractionation

SEC fractionation was carried out on an Agilent 1290 Infinity LC with an Agilent Infinity diode array detector using the same conditions as described for the SEC-ICPMS experiments. Biological replicates for each condition were concentration-matched, then equal volumes of replicates were pooled together. A total of 95 μg protein was injected onto an Agilent Bio-SEC 3 column (3 μm , 300 \AA , 4.6 \times 300 mm) and allowed to elute for 25 min at a flow rate of 0.4 mL/min. Beginning at 4 min post-injection, 1-min fractions were collected for 10 min. Collected fractions were freeze-dried for subsequent protein identification. Protein standards (cytochrome *c*, myoglobin, bovine serum albumin, γ -globulin, and glutamate dehydrogenase) were run in the same fashion as experimental samples. Their size exclusion elution, observed at 280 nm, was used as a molecular weight marker to generate estimated weight ranges for SEC fractions collected.

Proteomics of SEC fractions

The SEC fractions of the proteomes were freeze-dried prior to analysis by shotgun proteomics. The fractionated proteome samples (fractions 1–6) from both FeS_2 and Fe(II)/HS^- growth conditions were reduced, alkylated, and cleaned by chloroform/methanol extraction prior to digestion with sequencing grade modified porcine trypsin (Promega). Tryptic peptides were then separated by reverse phase XSelect CSH C18 2.5 μm resin (Waters) on an in-line 150 \times 0.075 mm column using an UltiMate 3000 RSLCnano system (Thermo). Peptides were eluted using a 60-min gradient from 98:2 to 65:35 solution A:B ratio with solution A being 0.1% formic acid and 0.5% acetonitrile and solution B being 0.1% formic acid and 99.9% acetonitrile. Eluted peptides were ionized by electrospray (2.4 kV) followed by mass spectrometric analysis on an Orbitrap Eclipse Tribrid mass spectrometer (Thermo). MS data were acquired using the FTMS analyzer in profile mode at a resolution of 120,000 over a range of 375 to 1,400 m/z with advanced peak determination. Following HCD activation, MS/MS data were acquired using the ion trap analyzer in centroid mode and normal mass range with a normalized collision energy of 30%. Proteins were identified by database search using MaxQuant (Max Planck Institute) with a parent ion tolerance of 3 ppm and a fragment ion tolerance of 0.5 Da. Scaffold Q + S (Proteome Software) was used to verify MS/MS-based peptide and protein identifications. Protein identifications were accepted if they could be established with less than 1.0% false discovery and contained at least two identified peptides. Protein intensities were determined by the sum of the top 3 peptides of each protein. Protein probabilities were assigned by the Protein Prophet algorithm (30). Ratios of protein abundance in each SEC fraction were calculated between growth conditions based upon protein intensities (Table S1).

Identifications of metalloproteins

Identified proteins were submitted to the online functional annotation tool (DAVID) (31, 32) to generate functional annotations, gene ontology terms, ligand information, and protein domain classifications. The DAVID output file was searched for metal-related key words (i.e., iron, 4Fe-4S, zinc, molybdopterin, etc.) (Table S2). The intensities of predicted metal proteins from each fraction were then plotted to show protein abundance as a function of elution time (Tables S3 to S8).

Gene synteny analysis of [Fe-S] cluster assembly system

Gene synteny was investigated using the *M. barkeri* Fusaro database in MetaCyc (33). A comparative gene synteny analysis was then conducted for other *Methanosarcina* species. For this, we used *Methanosarcina* sp. Kolksee and *Methanosarcina acetivorans* C2A gene databases on MetaCyc and additionally scanned the *Methanosarcina barkeri* MS genome (NCBI RefSeq assembly [GCF_000970025.1](https://www.ncbi.nlm.nih.gov/assembly/GCF_000970025.1)) on the National Center for Biotechnology Information website (34).

RESULTS

Growth with different sources of Fe and S

Growth curves of *M. barkeri* Fusaro provided with Fe(II)/HS⁻ or FeS₂ as the sole Fe and S sources were generated to establish a baseline for growth under different mineral conditions. Analysis of the production of CH₄ and DNA (proxy for growth) shows that cells grew similarly between the two conditions, although growth was slightly better on Fe(II)/HS⁻ (Fig. S1a and b) as observed previously (19). This is likely due to the requirement that cells reduce FeS₂ prior to assimilation, which would require a diversion of electrons to the mineral surface (19). Total sulfide (H₂S/HS⁻/S²⁻) was calculated from the measured aqueous concentrations using Henry's law to account for headspace sulfide and found that the maximum amount of total sulfide released from the mineral was 1.04 to 1.31 μmol. Aqueous HS⁻ concentrations ranged from 0.73 μM to 9.71 μM with a maximum concentration observed at 4 days post-inoculation (Fig. S1c), consistent with reductive dissolution of FeS₂. Past experiments with FeS₂ under these incubation conditions have demonstrated no HS⁻ production in the absence of cells (19, 20). With these data in hand, the experiment was repeated to generate biomass for proteomic analysis with cells harvested at mid-log phase (i.e., 4 days).

Fe and S source alters the metalloproteome

The soluble protein fractions of *M. barkeri* Fusaro grown with Fe(II)/HS⁻ or FeS₂ were analyzed using SEC-ICPMS. The elution profiles of five metals critical for methanogens (⁵⁶Fe, ⁵⁹Co, ⁶⁰Ni, ⁶⁶Zn, and ⁹⁵Mo) were monitored by SEC-ICPMS and this revealed a wide distribution of metals putatively associated with proteins and protein complexes of low and high molecular weight in both conditions (Fig. 1; Fig. S2). Furthermore, this indicated that the protocol was capable of maintaining native protein and protein complex structures.

Next, the effect of different sources of Fe and S on putative Fe metalloproteins was investigated. There was an increased Fe signal in the FeS₂ condition throughout the entire elution profile (Fig. 1). The majority of the protein-bound Fe, in this condition, eluted early (4–5.5 min) (>300 kDa based on molecular weight standards). In contrast, Fe(II)/HS⁻ grown cells had a low-intensity Fe signal in this high molecular weight region with the majority of the protein-bound Fe eluting after 7 min. The large Fe peak was absent in abiotic control sample mimicking the FeS₂ growth condition (Fig. S3). SEC-ICPMS analysis of Fe(II)/HS⁻-grown cells exposed to FeS₂ before harvesting also lacked the large Fe signal observed in the FeS₂ growth condition. Together, the results show that the Fe present in the high molecular weight region of SEC elution profiles in cells grown on FeS₂ is a biological consequence of culturing these cells on the mineral substrate.

Next, protein-associated Co, Ni, and Zn were investigated (Fig. 1). In the FeS₂ condition, each of these metals had a low-intensity signal in the 4- to 6-min region which coincided with the large Fe peak. These metals had a large peak between 6 and 8 min. Cobalt and Ni also had broad maxima spanning the 8- to 11-min range that were absent in the Zn profile. In the Fe(II)/HS⁻-grown cells, the Co, Ni, and Zn profiles also looked very similar. Each signal increased in intensity around 6 min and plateaued until approximately 10 min for Co and Ni, and 8 min for Zn. The SEC-ICPMS metal profiles for FeS₂-grown cells and FeS₂-grown cells incubated with Fe(II)/HS⁻ were nearly identical (Fig. S4). The data demonstrate the metalloproteome from the FeS₂ condition has more Co, Ni,

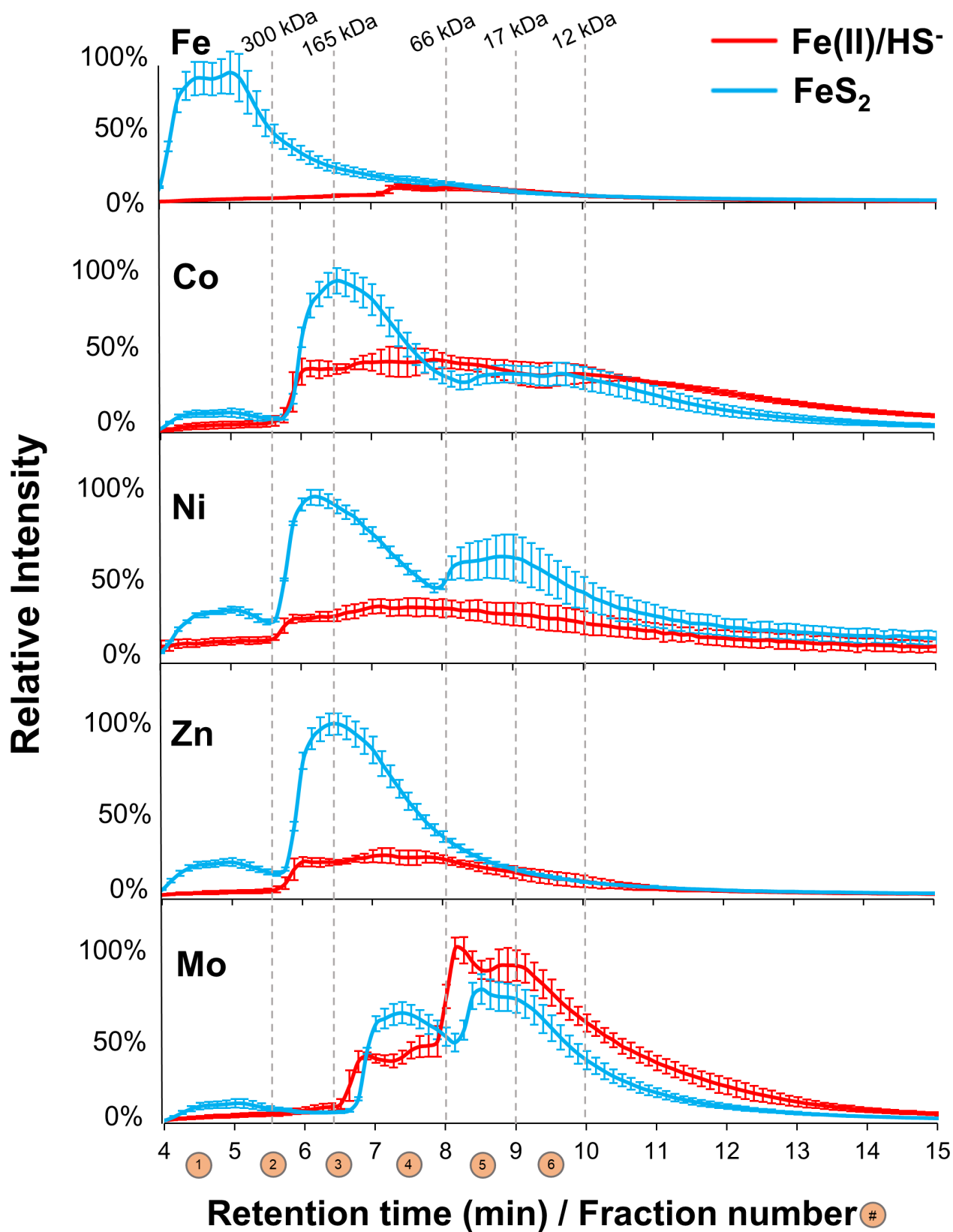


FIG 1 *Methanosarcina barkeri* metalloproteome response to Fe and sulfur (S) growth conditions. *M. barkeri* was grown on soluble ferrous iron (Fe(II)) and HS⁻ (red) or insoluble FeS₂ (blue). The soluble proteomes from both culture conditions were analyzed by SEC-ICPMS. The traces show the size distribution and intensity of protein-bound Fe, Co, Ni, Zn, and Mo. Error bars are ±1 standard deviation. SEC fractions are indicated by circles. Column void volume is at 4 min.

and Zn than the Fe(II)/HS⁻ condition, and that this was not caused by residual HS⁻ during sample preparation.

The final metal analyzed was Mo (Fig. 1). The elution profiles for FeS₂ and Fe(II)/HS⁻ growth conditions were similar. Both growth conditions had a significant increase in Mo before 7 min, followed by multiple maxima through 11 min. A larger maxima was observed in the FeS₂ condition between 7 and 8 min, while the peaks after 8 min were higher in the Fe(II)/HS⁻ condition.

Fe and S source impacts metal uptake

The initial SEC-ICPMS analysis revealed differences in the metalloproteome between growth conditions. We used ICPMS to quantify total intracellular metal content (relative to protein concentration) to address whether these changes were the result of metal redistribution or a consequence of metal uptake. There was significantly more Fe, Ni, and Zn in the FeS₂ condition (P -value <0.05), but no significant difference in the Co abundance between conditions (Fig. S5). Interestingly, there was significantly more Mo in the Fe(II)/HS⁻ condition (P -value <0.05).

Sulfide alters metal solubility and availability

To assess differences in the availability of trace metals between growth conditions, abiotic reactors were set up with soluble Fe(II)/HS⁻ or with FeS₂ and analyzed with ICPMS (Fig. 2). Reactors containing Fe(II) without a S source served as a control. Comparison of the Fe(II)/HS⁻ samples with the control samples at the start of the experiment showed a significant decrease ($P = 0.05$) in Fe, Co, and Ni concentrations. Zn was undetected in the Fe(II)/HS⁻ samples, yet was detected in the control and FeS₂ samples. During this initial timepoint, only Fe and Co were significantly lower ($P = 0.05$) in the FeS₂ condition than the control condition. At this initial timepoint, Co and Ni were significantly higher ($P = 0.05$) in the FeS₂ condition than in the Fe(II)/HS⁻ condition. Zn was also detected in the FeS₂ condition. Meanwhile, both conditions saw additional changes after 4 days of incubation. Mo became significantly lower in the Fe(II)/HS⁻ condition than both the control and the FeS₂ condition. Ni concentrations were significantly higher in the Fe(II)/HS⁻ condition than in the FeS₂ condition, but both were much lower than the control. In both the Fe(II)/HS⁻ samples and the FeS₂ samples, Co was undetected in at least one of the replicates.

Identification of metalloproteins through shotgun proteomics

A total of 1,563 proteins [out of 3,698 open reading frames in the *M. barkeri* Fusaro genome (35)] were identified across all SEC fractions (Table S1). DAVID analysis returned information on the metal binding potential for 1,409 proteins (Table S2). One hundred forty-seven Fe proteins, 30 Ni proteins, 18 Mo proteins, 84 Zn proteins, and 25 Co proteins were identified in the SEC fractions collected (Tables S3 to S7). The results are broken out into target groups with predicted or characterized functions below.

Proteins putatively involved in Fe storage and trafficking in FeS₂-grown cells

To identify the protein responsible for the large molecular weight iron peak, we examined the fold-change (FeS₂:Fe(II)/HS⁻) of the proteins identified in fraction 1. Only two of the identified proteins had changed over twofold: subunit K of A1AO H⁺ ATPase Mbar_A0390 and the uncharacterized Mbar_A0933 (Table S1). BLAST analysis of Mbar_A0933 suggests it is a HEAT domain-containing protein. Neither protein contained canonical Fe-binding domains or motifs. The characterized Fe-binding proteins in fraction 1 did not have a significant change in abundance. We speculated that *M. barkeri* could store Fe as a thioferrate-like mineral in a large protein complex, similar to the IssA protein in *Pyrococcus furiosus* which lacks classical Fe-binding domains (36). *M. barkeri* Fusaro encodes two IssA homologs, Mbar_A1833 and Mbar_A1834. Both homologs were more abundant in the Fe(II)/HS⁻ condition (Fig. S6) and the elution time from SEC matched the expected protein size.

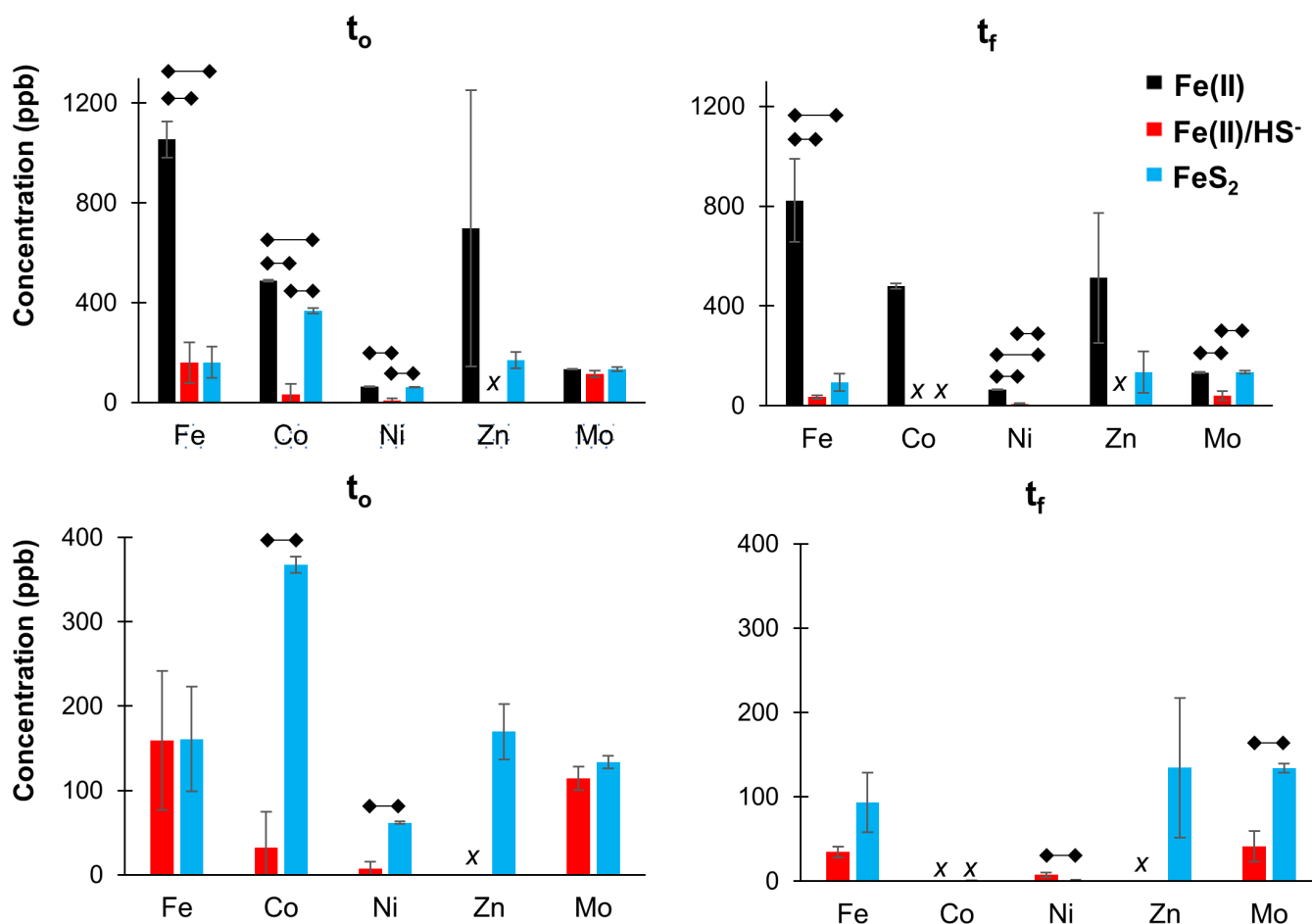


FIG 2 Influence of ferrous iron (Fe(II)), Fe(II) and HS⁻, or FeS₂ on the availability of Fe, Co, Ni, Zn, and Mo. Metal concentrations from abiotic reactors were analyzed immediately after setting up the experiment (left, t₀) and after 4 days (right, t_f) for media containing trace metals and varying sources of iron and sulfur; soluble Fe(II) and no sulfur (black), soluble Fe(II)/HS⁻ (red), and mineral FeS₂ (blue). The bottom row is the same as the top row without displaying the soluble Fe(II) with no sulfur data for clarity. ◆◆ indicates significant difference at *P*-value = 0.05. x indicates one or more samples were below the limit of detection.

Analysis of fractions 2–6 revealed the presence of two Fe-containing proteins that had an intensity ratio greater than fivefold (FeS₂:Fe(II)/HS⁻) and shared a similar elution pattern across the sampled fractions (Fig. 3a). These proteins were initially identified as a cysteine desulfurase (Nif-S) homolog and an iron-sulfur cluster assembly protein (Nif-U) homolog. The genes for these two proteins reside next to each other on the *M. barkeri* Fusaro genome (Mbar_A2424 and Mbar_A2425). After examining the genomic context, it became evident that they could also be categorized in the recently described minimal [Fe-S] cluster assembly pathway as Mis-S and Mis-U, respectively (25). Bioinformatic analysis of the cysteine desulfurases of *M. barkeri* Fusaro, *M. acetivorans* C2A, and *Methanosarcina* sp. Kolksee revealed gene synteny with the MIS protein localization (Fig. 3b). The minimal [Fe-S] cluster mechanism SufCB was also identified in our data set. Both SufC (Mbar_A1078) and SufB (Mbar_A1079) were more abundant in the Fe(II)/HS⁻ condition (Fig. 4a). Interestingly, the elution patterns of these proteins were not uniform, as was observed for the MIS proteins.

Proteins involved in cysteine biosynthesis

The MIS genes that were upregulated in the FeS₂ condition are located between cysteine biosynthesis genes (Mbar_A2421, Mbar_A2422, and Mbar_A2427), a transcriptional regulator (Mbar_2423), and a hypothetical protein that was determined to be *nada*

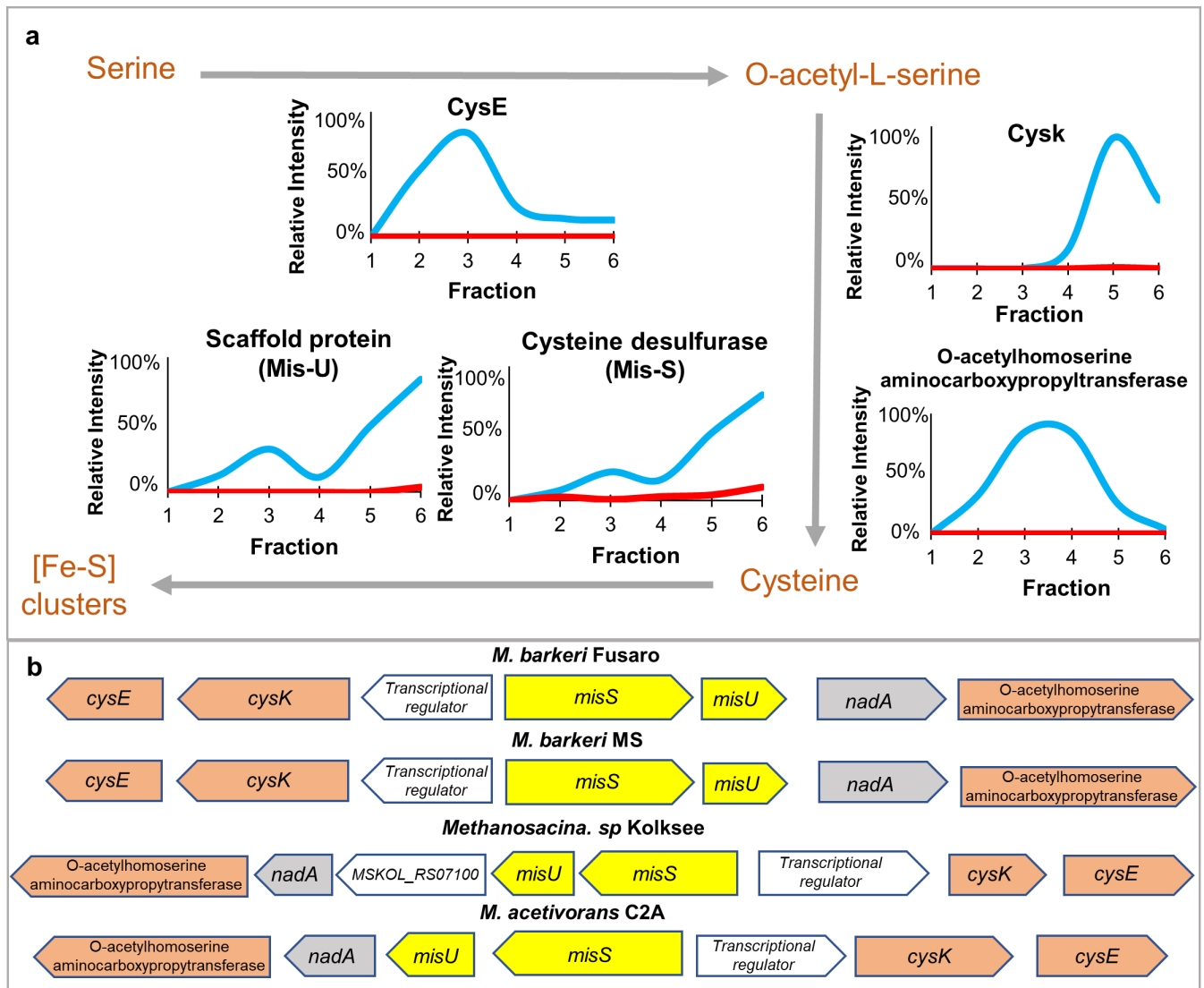


FIG 3 Fate of iron and sulfur in FeS₂-grown *Methanosarcina barkeri* cells. (a) Protein intensity of size exclusion chromatography fractions show an upregulation in cysteine biosynthesis and [Fe-S] cluster assembly pathway proteins in FeS₂-grown cells (blue) when compared to the Fe(II) and HS⁻ condition (red). Relative molecular weight ranges based upon standards; fraction 1 (460–970 kDa), fraction 2 (220–460 kDa), fraction 3 (100–220 kDa), fraction 4 (49–100 kDa), fraction 5 (23–49 kDa), and fraction 6 (11–23 kDa). (b) Analysis of the genomic context of cysteine desulfurase (MisS) and the scaffold protein (MisU) in *Methanosarcina barkeri* Fusaro, *Methanosarcina barkeri* MS, *Methanosarcina* sp. Kolksee, and *Methanosarcina acetivorans* C2A reveals gene synteny. Genes and genomes are adapted from Biocyc and NCBI databases and are not scaled to size. Conserved genes are color coordinated. The MIS system is in yellow, cysteine biosynthesis in salmon, and *nadA* in gray.

(Mbar_RS12880), which encodes for quinolate synthase A. The cysteine biosynthesis proteins were absent or in low abundance in the Fe(II)/HS⁻ fractions (Fig. 3a). These data show that under FeS₂ growth conditions, *M. barkeri* cysteine biosynthesis proteins and MIS proteins are more abundant. The localization of these genes in *M. barkeri* and the gene synteny across other *Methanosarcina* species suggests that these proteins work together in [Fe-S] cluster biosynthesis and respond to Fe availability. O-phospho-L-seryl-tRNA^{Cys}-tRNA synthase (Mbar_A1638), an enzyme involved in the alternative tRNA-dependent cysteine biosynthesis pathway, was present in our data and was more abundant in the Fe(II)/HS⁻ condition (Fig. 4b). While this enzyme was present in the FeS₂ growth condition, the elution profiles were different between conditions. In the Fe(II)/HS⁻ growth condition, it was bimodal with maxima in fractions 2 and 5, while in the FeS₂ condition, it was monomodal and centered between fractions 3 and 4. We observed

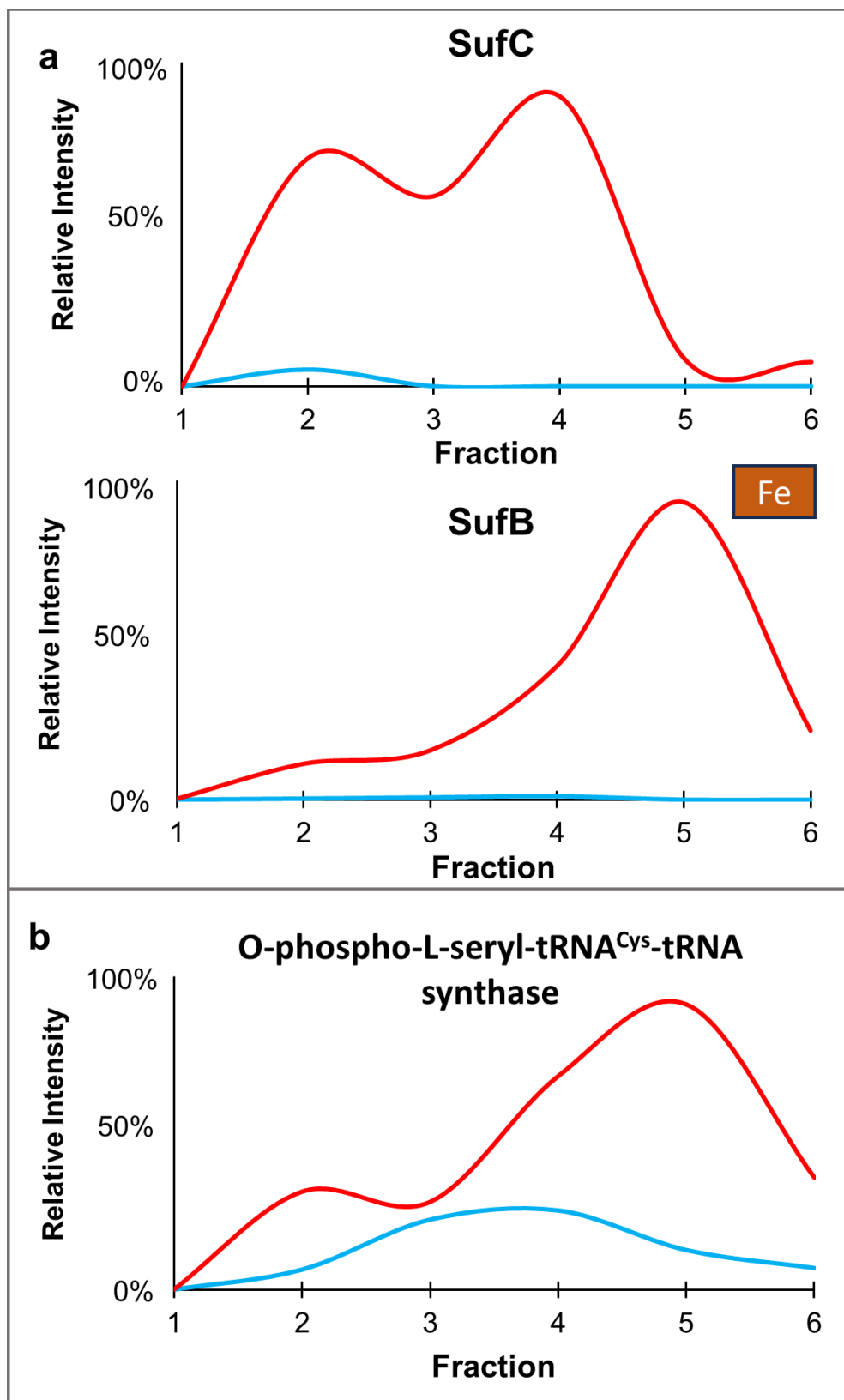


FIG 4 Influence of iron and sulfur source on iron-sulfur cluster assembly and cysteine biosynthesis protein intensity in *Methanosarcina barkeri* cells. Proteomics was done on individual size exclusion chromatography fractions for Fe₂ (blue)- or (Continued on next page)

FIG 4 (Continued)

Fe(II)/HS⁻ (red)-grown cells. Intensities of the identified proteins were plotted from each fraction for (a) Suf proteins and (b) O-phospho-L-seryl-tRNA^{Cys}-tRNA synthase. Intensities were the average of the top 3 precursor ion intensities. Molecular weight ranges for each fraction: fraction 1 (460–970 kDa), fraction 2 (220–460 kDa), fraction 3 (100–220 kDa), fraction 4 (49–100 kDa), fraction 5 (23–49 kDa), and fraction 6 (11–23 kDa). Metal cofactors derived from DAVID functional annotation are given.

three other instances of cysteine desulfurases in the *M. barkeri* genome (Fig. S7). Two of these enzymes were not detected (Mbar_A3201, and Mbar_A1737) in the proteomics data (Fig. S7a). The other cysteine desulfurase (Mbar_A0729) was detected in equal abundance between conditions (Fig. S7b).

Proteins involved in methanogenesis and electron transport chains

The three subunits of MCR, alpha (Mbar_A0893, which contains nickel tetrapyrrole F₄₃₀), beta (Mbar_A0897), and gamma (Mbar_A0894), were detected in our proteomics data. All three subunits were more abundant in the FeS₂ growth condition and eluted with a unimodal SEC profile consistent with an assembled complex (Fig. 5a). The *mcr* locus encodes for two other Mcr proteins that are denoted McrC (Mbar_A0895) and McrD (Mbar_A0896). These two proteins had the opposite behavior in comparison to McrABG and were more abundant in the Fe(II)/HS⁻ condition. The bimodal elution profile indicates different oligomeric states (Fig. 5a). The change in Mcr inspired investigation of other core methanogenesis proteins. The subunits of coenzyme F₄₂₀-reducing [NiFe]-hydrogenase [alpha (Mbar_A0452), beta (Mbar_A0449), and gamma (Mbar_A0450)] were more abundant in the FeS₂ condition (Fig. 5b). Five corrinoid proteins (Mbar_A0945, Mbar_A3634, Mbar_A0842, Mbar_A740, Mbar_A1063), which are involved in the methylotrophic methanogenesis pathway, were identified in our data. Four of these corrinoid proteins were more abundant in the FeS₂ condition (Fig. 5c). These elution profiles show that the F₄₂₀-reducing [NiFe]-hydrogenase and most of the corrinoid proteins are higher in abundance in the FeS₂ condition.

A premise of this work was to maintain the native state of metalloprotein complexes. Therefore, our cell lysis and protein extraction methods were very gentle. This allowed us to retain several membrane protein complexes in the protein fraction. Among these were several membrane metalloproteins associated with methanogenesis and these were differentially expressed based on the growth condition. Methyltetrahydromethanopterin:coenzyme M methyltransferase (Mtr) is a membrane-bound complex that generates the ion motive force and the formation of methyl-CoM. Mtr is composed of eight subunits (MtrA-H) (37, 38). Six of these subunits were identified through the fractions: MtrA (Mbar_A1258 which binds Co), MtrB (Mbar_A1259), MtrC (Mbar_A1260), MtrE (Mbar_A1262), MtrG (Mbar_A1256), and MtrH (Mbar_A1255). The abundance of these subunits revealed that the stoichiometry of the complex was condition-dependent (Fig. 6a). In the Fe(II)/HS⁻ condition, there was a distinct bimodal distribution with maxima around fractions 2 and 4. In contrast, the distribution of these subunits in FeS₂ conditions was typically more broad with a larger single peak.

The elution profiles of energy-conserving [NiFe]-hydrogenase (Ech) show a change in both abundance and elution profile between culture conditions (Fig. 6b). In the Fe(II)/HS⁻ condition, there was a large Ech complex centered on fraction 2 and a smaller complex in fractions 4 and 5. EchB (Mbar_A0151) was the only subunit that had increased intensity in the smaller complex in the Fe(II)/HS⁻ condition. In the FeS₂ condition, all of the subunits, except EchA (Mbar_A0152) and EchC (Mbar_A0150), were more abundant in the smaller complex (fractions 4 and 5). In all cases, except for EchC and EchE (Mbar_A0148), protein abundance was higher in the Fe(II)/HS⁻ condition. The signal intensities for subunit EchC were very similar, while EchE was bimodal in the Fe(II)/HS⁻ condition with peaks in fractions 2 and 5 while being unimodal in the FeS₂ condition with a peak in fraction 4.

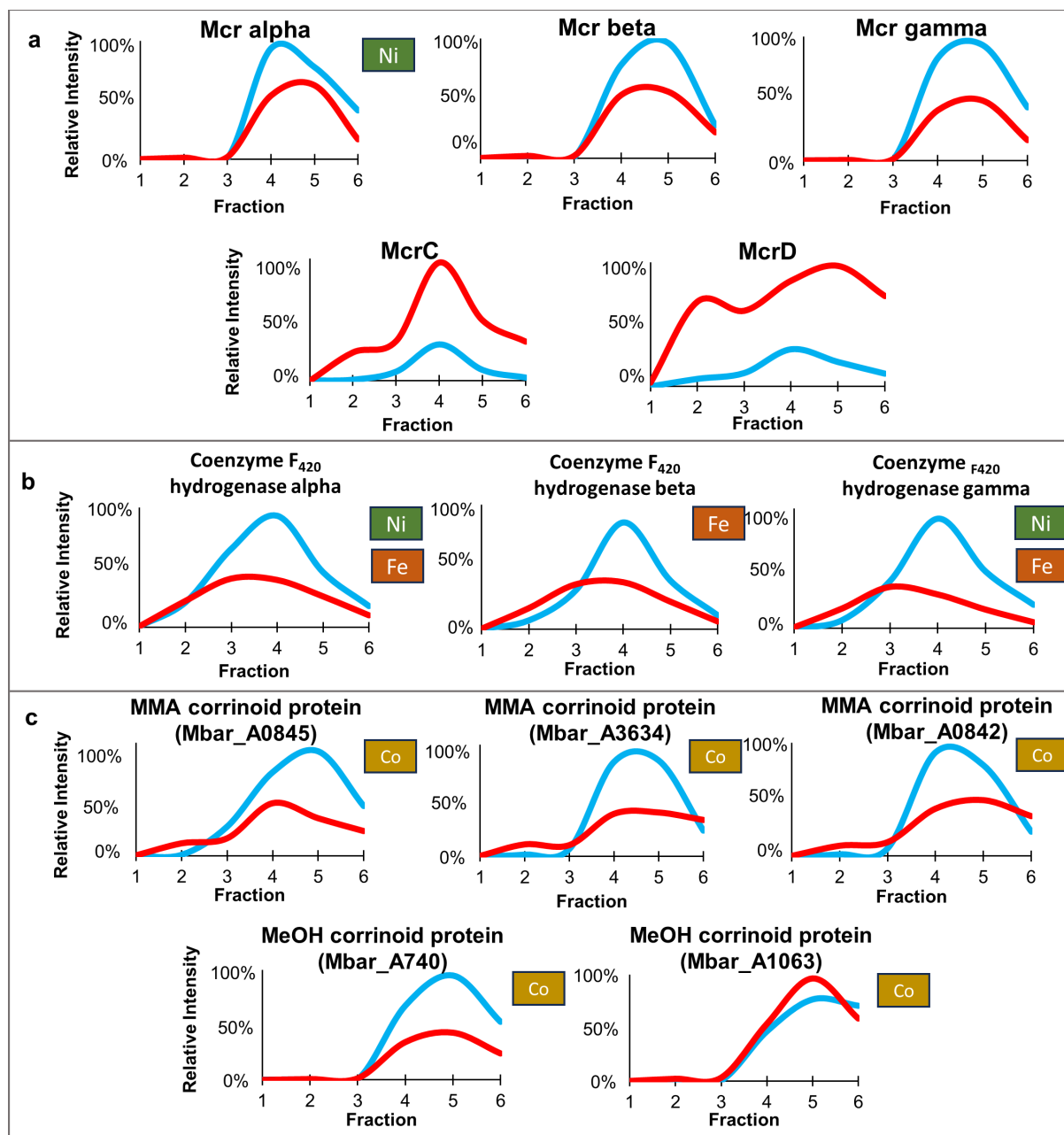


FIG 5 Source of iron and sulfur impacts methanogenesis protein intensities in SEC fractions. Proteomics was done on individual SEC fractions for FeS₂ (blue)- and Fe(II)/HS⁻ (red)-grown cells. Intensities of the identified soluble methanogenesis proteins were plotted from each fraction for (a) Mcr subunits, (b) F₄₂₀-reducing [NiFe]-hydrogenase subunits, and (c) corrinoid methyltransferase proteins. Locus tags are given for proteins. Monomethylamine is abbreviated as MMA and methanol is abbreviated as MeOH. Intensities were the average of the top 3 precursor ion intensities. Relative molecular weight ranges based upon standards: fraction 1 (460–970 kDa), fraction 2 (220–460 kDa), fraction 3 (100–220 kDa), fraction 4 (49–100 kDa), fraction 5 (23–49 kDa), and fraction 6 (11–23 kDa). Metal cofactors derived from DAVID functional annotation are given.

The elution patterns of the methanophenazine-reducing [NiFe]-hydrogenase (Vht) subunits were similar between conditions (Fig. 6c). There were increased intensities for the Vht large (Mbar_A1846) and small subunit (Mbar_A1847) in the Fe(II)/HS⁻ condition, but similar intensities for the cytochrome *b* subunit (Mbar_A1845). Heterodisulfide reductase (Hdr) subunits HdrD (Mbar_A1599) and HdrE (Mbar_A1598) were more abundant in the Fe(II)/HS⁻ condition (Fig. 6d). HdrD was bimodal in the Fe(II)/HS⁻ condition with peaks around fraction 2 and 4 but had a single peak in the FeS₂ condition

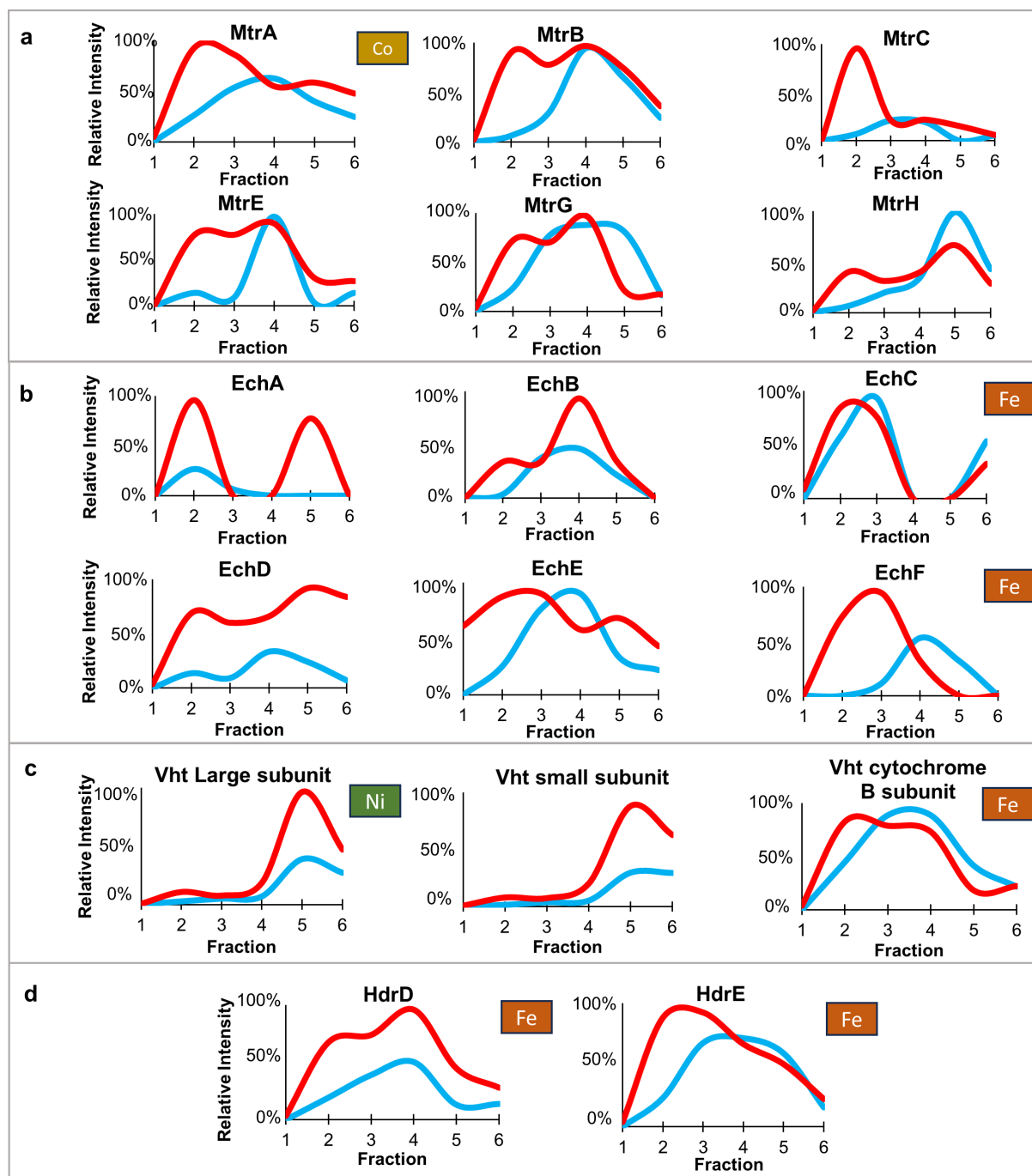


FIG 6 Iron and sulfur source alters membrane methanogenesis and electron transport chain proteins in SEC fractions. Proteomics was done on individual SEC fractions for FeS_2 (blue)- and Fe(II)/HS^- (red)-grown cells. Intensities of the identified membrane methanogenesis and electron transport proteins were plotted from each fraction for (a) Mtr subunits, (b) Ech subunits, (c) Vht subunits, and (d) Hdr subunits. Intensities were the average of the top 3 precursor ion intensities. Relative molecular weight ranges based upon standards: fraction 1 (460–970 kDa), fraction 2 (220–460 kDa), fraction 3 (100–220 kDa), fraction 4 (49–100 kDa), fraction 5 (23–49 kDa), and fraction 6 (11–23 kDa). Metal cofactors derived from DAVID functional annotation are given.

centered around fraction 4. In the FeS_2 condition, there was only one peak centered around fraction 4. HdrE had a large peak in fraction 2 in the Fe(II)/HS^- condition with peak tailing through the rest of the elution. In the FeS_2 condition, there was a broad peak centered around fraction 4. The elution profiles for these proteins show that the oligomeric state and abundance change for some of the proteins depending on the growth condition.

Fe and S source affects ribosomal proteins

Five of the ribosomal proteins in the data set were annotated as Zn-containing proteins; three 50S proteins L24e (Mbar_A3386), L37Ae (Mbar_A2508), and L44e (Mbar_A1569), and two 30S proteins, S14 type Z (Mbar_A0096) and S27e (Mbar_A1568). These proteins have been shown to bind Zn in other organisms (39–42). The SEC distribution of these proteins differed between conditions (Fig. S8). L37Ae and S27e had a bimodal profile in the Fe(II)/HS⁻ condition but were unimodal in the FeS₂ condition. L44e, however, was unimodal in both conditions, but was present in earlier eluting fractions in the Fe(II)/HS⁻ condition, indicative of being part of a larger complex. L24e showed a similar distribution through the fractions between conditions, but was much higher in the Fe(II)/HS⁻ condition while being greatly reduced in the FeS₂ condition. S14 type Z showed a bimodal distribution in the Fe(II)/HS⁻ condition but is absent in the FeS₂ condition. There was also a change in distribution of ribonuclease J (Mbar_A1418), which also binds Zn. The elution profiles of these proteins show that the Fe and S source affects ribosomal proteins and protein complexes.

DISCUSSION

Methanogens commonly inhabit euxinic environments, conditions that favor the precipitation of many transition metals as sulfide minerals including Fe as FeS and FeS₂ (43). The recent discovery that *M. barkeri* and other methanogens can meet Fe and S demands directly from the minerals FeS and FeS₂ initiated a paradigm shift in how we think about metal availability in the environment and their acquisition in methanogens, and other organisms that inhabit euxinic environments (19, 20, 24). Our study aimed to characterize the difference in the metalloproteome of *M. barkeri* when grown with Fe(II)/HS⁻ (mimicking euxinic conditions) versus FeS₂. Importantly, in both conditions, Fe/S is in excess of cellular demands—potentially mitigating possible dosage effects due to the different availabilities of Fe and S. The concentrations used were based on past work (19, 20) that showed broad phenotypic changes in cells grown on Fe(II)/HS⁻ and FeS₂ so that detected changes in the metalloproteome could expand on these phenotypes. We leveraged an approach that profiled the metalloproteome while retaining the native structure of protein complexes using SEC coupled with ICPMS. SEC is a powerful tool for fractionating the proteome while maintaining complexes in the native state (Fig. S2) (44). Our data clearly show a restructuring of the metalloproteome in terms of metal distribution (Fig. 1), abundance of the metalloproteins involved in methanogenesis (Fig. 5), electron transport (Fig. 6), and ribosomal processes (Fig. S8) based on the source Fe and S provided. Furthermore, Fe(II)/HS⁻ and FeS₂ on their own were shown to alter the soluble metal pool (Fig. 2).

Metal availability and the structure of the metalloproteome are inherently linked (45, 46). Our data show that intracellular metal concentrations are dependent on the form of Fe and S that are available (Fig. S5). This appeared to be due in part to differential solubility of metals in the presence of Fe(II)/HS⁻ or FeS₂, independent of any biological mechanism. In abiotic reactors, soluble metal concentrations diminished in both conditions over a period of 4 days incubation. In the HS⁻ condition, this was due to the formation of metal sulfide precipitates, which were visually apparent and black in color, whereas in the FeS₂ condition, it was likely due to adsorption on the FeS₂ mineral (Fig. 2) (15–18). This finding implies that extra care must be taken when interpreting previous experimental regimes using Fe(II)/HS⁻ and metal accumulation in cells (47). The consequence of these different abiotic mechanisms, precipitation in the sulfide condition and adsorption in the FeS₂ condition, likely changes Co, Zn, Ni, and Mo acquisition strategies. This may facilitate acquisition of metals in the FeS₂ condition because they are colocalized with the FeS₂ surface, if not soluble (as was the case for Zn and Mo). This is supported by increased metal uptake of Fe, Ni, and Zn in the FeS₂-grown cells (Fig. S5) and increased intensities of these metals observed in the SEC-ICPMS data (Fig. 1). Cytosolic Co, however, was similar between conditions (Fig. S5), and abiotic reactors

showed this element fully precipitated out of solution for both sources of Fe and S (Fig. 2). The observed changes in cytosolic metal concentrations and the metalloproteome are likely due to a dynamic multiplex response of metal acquisition and metabolic strategies. Indeed, large proteomic changes involving amino acid, protein, nucleic acid, and central carbon metabolism have been observed in *M. voltae* and *M. barkeri* MS in these same growth conditions leading to distinct phenotypes (21, 22).

Ribosomal protein expression was a major factor of phenotypic growth differences in *M. barkeri* MS grown under differing Fe and S sources (including FeS₂ and Fe(II)/HS⁻) (21). It was speculated that this could be due to either differing growth rate kinetics between conditions or that these ribosomal proteins have moonlighting functions, performing non-ribosomal functions. In the FeS₂ growth experiments, *M. barkeri* Fusaro appears to use the recently proposed MIS cluster assembly to synthesize [Fe-S] clusters, as they were the only [Fe-S] cluster biosynthesis proteins in high abundance (Fig. 3a). The MIS consists of a cysteine desulfurase and a scaffold protein, but apparently no ATPase (25). The presence of these two proteins is significant because cysteine is the putative S source for the [Fe-S] clusters in this pathway (25). In contrast, the sulfidic growth experiments suggest that *M. barkeri* Fusaro uses the SufCB pathway for [Fe-S] cluster synthesis with the tRNA-dependent cysteine biosynthesis pathway (Fig. 4). The sulfur source for SufCB pathway has yet to be elucidated but could be FeS_{aq} clusters (either intact or dissolved), as has been recently proposed (14).

The different elution patterns between SufCB may be indicative of another protein involved in [Fe-S] cluster synthesis, potentially providing the unidentified sulfur source. Indeed, recent work investigating the role of these “minimal SUF” systems in a closely related strain, *Methanosarcina acetivorans*, has demonstrated they are not required for growth when HS⁻ or cysteine are provided as S sources potentially because these organisms also encode components of the ISC [Fe-S] biosynthetic system (48). Interestingly, the O-phospho-L-seryl-tRNA^{Cys}-tRNA synthase elution pattern has striking similarities to both SufC and SufB elution patterns (Fig. 4). This would provide a plausible explanation as to why *M. barkeri* uses the tRNA-dependent cysteine biosynthesis pathway with SufCB instead of the cysteine biosynthesis pathway used under FeS₂ growth conditions.

Based on previous results with *M. voltae* (14), we speculated that the large SEC-ICPMS Fe peak herein may be excess Fe that was being stored as thioferrate or a thioferrate-like mineral in supramolecular-IssA homolog structures to mitigate Fe toxicity (36). The two IssA homologs in *M. barkeri*, however, were low in abundance or not detected in the FeS₂ condition, yet were very abundant in the Fe(II)/HS⁻ condition (Fig. S6), suggesting they are not involved in coordination of this putative mineral in *M. barkeri*. This does not rule out the presence of thioferrate or a thioferrate-like mineral in the FeS₂ cells and may point to other functions of IssA homologs in *M. barkeri*. Alternatively, the IssA homologs could be post-translationally modified and therefore go undetected by the proteomics workflow used here. Another possibility is that one or both of the homologs may be tightly associated to the presumed thioferrate-like mineral, or binding in an orientation, that is not amenable to tryptic digestion.

The genetic differences between *M. voltae* and *M. barkeri* should not be ignored in the present interpretations. *M. voltae* is a type I methanogen and lacks homologs of cysteine desulfurase, while *M. barkeri* is a type II methanogen and has multiple homologs of cysteine desulfurase (49, 50). The ability of *M. barkeri* to metabolize cysteine to form [Fe-S] clusters may mitigate a need for Fe storage using IssA homologs. Our data, in the context of previous findings (14), suggest that Fe and S are trafficked differently in *M. barkeri* than in *M. voltae*, which may be due in part to the presence of a cysteine desulfurase and additional [Fe-S] biosynthetic machinery in the former organism.

A striking difference observed in our data was the expression of Mcr proteins. While there was higher abundance of the catalytic MCR subunits alpha, beta, and gamma in the FeS₂ condition, the McrC and McrD subunits were much higher in the Fe(II)/HS⁻ condition (Fig. 5a). There is little in the literature on the roles of McrC and McrD. McrC is

necessary for activating MCR in *Methanothermobacter marburgensis* (51), and McrD has been proposed to be involved in MCR folding and cofactor insertion (52, 53). Since these proteins are working for the collective goal of supporting MCR folding and activation, we hypothesize that McrC and McrD are overexpressed in the Fe(II)/HS⁻ condition to compensate for the lower expression of MCR. Another possibility is that the species of Fe brought into cells when grown on FeS₂ versus those grown with HS⁻ differs and leads to different needs for these McrC and McrD proteins in MCR complex formation.

The findings of our research highlight the complex interplay between metal availability in the environment, cellular acquisition, metabolism, and metalloproteome composition. *M. barkeri* employs different metabolic strategies when grown under Fe(II)/HS⁻ and FeS₂ growth conditions. These metabolic strategies have significant effects on the metalloproteome and protein complexes involved in numerous essential functions including methanogenesis and [Fe-S] cluster biosynthesis. Our work also shows a connection between the MIS and cysteine biosynthesis pathway when *M. barkeri* is grown under FeS₂ conditions which are distinct from the SufCB and t-RNA-dependent cysteine biosynthesis pathways employed when grown under euxinic conditions. The consequence of this may influence Fe trafficking and storage. This work provides important and new insights into the mechanisms used by methanogens to acquire essential Fe and S and highlights exciting new areas for future investigation into metal and S acquisition under different environmental conditions. Additionally, the plasticity of the metalloproteome observed here should motivate additional study into the redistribution of the metalloproteome and stoichiometry of protein complexes of other microorganisms that occupy a multitude of diverse environment types where essential transition metals are expected to be differentially available.

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DATA AVAILABILITY

The identified proteins and their UniProt accession numbers and alternate identifiers are available in the supplemental tables. *M. barkeri* gene names for proteins of interest are given throughout the text.

ADDITIONAL FILES

The following material is available [online](#).

Supplemental Material

Supplemental figures (AEM00516-24-s0001.pdf). Figures S1 to S8.

Supplemental tables (AEM00516-24-s0002.xlsx). Tables S1 to S8.

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