

Reduced susceptibility of thin *Pseudomonas aeruginosa* biofilms to hydrogen peroxide and monochloramine

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W. L. COCHRAN, G. A. McFETERS AND P. S. STEWART. 2000. *Pseudomonas aeruginosa* attached to alginate gel beads in sparse, thin biofilms exhibited reduced susceptibility to monochloramine and hydrogen peroxide compared with planktonic cells of the same micro-organism. Disinfection rate coefficients for planktonic bacteria averaged $0.551 \text{ mg}^{-1} \text{ min}^{-1}$ for monochloramine and $3.1 \times 10^{-4} \text{ l mg}^{-1} \text{ min}^{-1}$ for hydrogen peroxide. The corresponding values for 24-h-old biofilm cells were $0.291 \text{ mg min}^{-1}$ and $9.2 \times 10^{-5} \text{ l mg}^{-1} \text{ min}^{-1}$ for monochloramine and hydrogen peroxide, respectively. Several pieces of evidence support the interpretation that the reduced susceptibility of biofilm was not due simply to inadequate delivery of the antimicrobial agent to the local environment of the attached cells. No correlation between biofilm susceptibility and biofilm initial areal cell density was observed. Rapid delivery of hydrogen peroxide to the attachment surface, and subsequently to the interior, of the alginate gel beads was visualized by a direct experimental technique. Theoretical analysis of unsteady diffusion and diffusion–reaction interactions also argued against any significant delay or barrier to antimicrobial or oxygen delivery. It was hypothesized that new genes are expressed when bacteria attach to a surface and begin to form a biofilm and that some of the resulting gene products reduce the susceptibility of the cell to antimicrobial agents including oxidative biocides such as monochloramine and hydrogen peroxide.

INTRODUCTION

When micro-organisms grow attached to a surface in the form of a biofilm they exhibit remarkable resistance to all types of antimicrobial challenge when compared with the same micro-organisms grown in suspended cultures (Gilbert and Brown 1995). This phenomenon of reduced biofilm susceptibility explains the chronic nature of biofilm infections and thwarts efforts to control biofouling in a wide variety of industrial settings (Costerton, Marrie and Cheng 1985). The fundamental physical, chemical and biological mechanisms by which biofilm micro-organisms escape killing by biocides and antibiotics are still incompletely understood.

We distinguish three types of hypothesized mechanisms of

reduced biofilm susceptibility to antimicrobial agents. The first of these is failure of the antimicrobial agent to penetrate the full depth of the biofilm. This does indeed occur in certain instances (de Beer, Srinivasan and Stewart 1994; Chen and Stewart 1996; Xu, Stewart and Chen 1996; Liu *et al.* 1998; Stewart, Grab and Diemer 1998). The reaction–diffusion interaction underlying poor penetration is now well enough understood that it can be mathematically modelled (Stewart and Raquepas 1995; Chen and Stewart 1996; Dibdin *et al.* 1996; Stewart 1996). Penetration failure is most viable as a resistance mechanism when dealing with thick biofilms and highly reactive antimicrobials.

A second mechanism of biofilm reduced susceptibility requires that at least some of the cells within a biofilm experience a nutrient limitation that causes them to enter a slow-growing or starved state (Brown, Allison and Gilbert 1988). Slow or non-growing cells have been shown to be less susceptible to a variety of antimicrobial agents when compared

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with cells grown in rich media at high specific growth rates. There is considerable experimental support for this mechanism (Gilbert and Brown 1995) and also for striking spatial heterogeneity in the physiological status of bacteria within relatively thick (100 μm) biofilms (Wentland *et al.* 1996; Xu *et al.* 1998).

The third mechanism of reduced biofilm susceptibility, which is more speculative than the preceding hypotheses, is that at least some of the cells in a biofilm adopt a distinct, and relatively protected, biofilm phenotype. This phenotype is not the result of a nutrient limitation. Whereas the first two mechanisms outlined above both involve a transport limitation, in the first case for the antimicrobial agent itself and in the second case for a nutrient, there is no requirement for any transport limitation in the 'resistant biofilm phenotype' hypothesis. The practical importance of this mechanism, should it prove true, is profound because it implies that reduced susceptibility of biofilm is genetically programmed. This would open the door to many novel approaches to biofilm control. It is the possibility of an intrinsically distinct biofilm phenotype that motivated the work reported in this paper.

A good experimental demonstration of the existence of a resistant biofilm phenotype requires that transport phenomena be eliminated as factors influencing biofilm susceptibility. We have addressed this requirement by using a model biofilm system in which bacteria were attached in sparse, thin films to the surface of highly hydrated alginate gel beads. These gel beads can be maintained in an aqueous suspension with stirring. Their well-defined spherical geometry allows mass transfer characteristics between the bulk fluid and the surface of the gel bead (where bacteria are located) to be accurately calculated. Because the gel beads are readily permeated by solutes, antimicrobial agents can be delivered to all aspects of the attached cell.

The experimental aim of this study was to evaluate the efficacy of two antimicrobial agents on thin biofilms grown in a system where biocide mass transport was not an issue. The antimicrobial effects of monochloramine and hydrogen peroxide were assessed in this novel biofilm system by comparing disinfection rates of planktonic and attached cells.

MATERIALS AND METHODS

Bacterial strains, culture and enumeration

All disinfection experiments used *Pseudomonas aeruginosa* PAO1 (Holloway 1955). Strains were cultured in modified R2A broth (American Public Health Association *et al.* 1995) without K_2HPO_4 and starch, and with the addition of 3.4 mmol l^{-1} $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$. Enumeration of bacteria was performed by serial dilution in phosphate-buffered saline (PBS; pH 7.5) (US Food and Drug Administration 1984) followed

by drop plating 10- μl drops (Miles and Misra 1938; Hoben and Somasegaran 1982) onto R2A plates (Difco Laboratories, Detroit, MI).

GFP expressing *Ps. aeruginosa*

To visualize bacterial cell distribution on the surface of alginate beads, a plasmid that constitutively expressed the green fluorescent protein (GFP) was introduced into *Ps. aeruginosa* PAO1. The gene for the GFP containing the *mut2* mutation (Cormick, Valdivia and Falkow 1996) was amplified from plasmid pBC*gfp* using PCR (Matthysse *et al.* 1996). The PCR primers used in the amplification were GFPSal3-5'GC GCGTCGACAGGAGAAGAAAAATGAGTAAACCA-GAAGA 3' and GFPHind4-5' GTACCTGGAATTCTA-CGAAGCTTATTTGTATAGTTCATCC 3'. The PCR product was digested with SalI and HindIII, and ligated into pUC19. The XbaI and HindIII fragment from pUC18, containing the *gfpmut2*, was then ligated into vector pMF36 (Franklin and Ohman 1993) behind the strong *trc* promoter, forming plasmid pMF230. Since pMF36 contains the *oriT* site and the stable replication fragment, it was mobilized into *Ps. aeruginosa* by triparental mating and was stably maintained. pMF36 does not contain *lacI*, resulting in constitutive expression of *gfp* from the *trc* promoter.

Alginate gel beads

The method of Smidsrød and Skjåk-Bræk (1990) was used to prepare gel beads. A 2% (w/v) sodium alginate solution was autoclaved for 30 min and then mixed on a stir plate overnight. Sterile 2-mm diameter beads were formed by dropping liquid alginate solution into stirred 1% (w/v) sterile $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$.

Biofilm formation

An overnight culture of *Ps. aeruginosa* was diluted to 10^6 colony forming units (c.f.u.) ml^{-1} in 500 ml of R2A in a 1500-ml beaker containing approximately 3000 alginate beads. Cells were allowed to attach to the beads for up to 2 h at room temperature with continuous stirring. The medium was decanted and replaced with sterile broth. Beads were incubated at room temperature for various time intervals from 3.25 to 72 h. Spent medium was removed and sterile R2A was added approximately every 4–8 h to maintain nutrient replete growth conditions. Before disinfection assays, beads were rinsed three times in R2A and incubated for an additional 3–4 h to ensure the absence of stationary phase cells.

Epifluorescence microscopy

GFP containing biofilms were grown for 24 and 48 h as described above. Micrographs of cells attached to alginate beads were taken using a Nikon optiphot microscope with a B2A epifluorescence filter, a Nikon N70 camera and Kodak Tmax 400 black and white film.

Preparation of disinfectants

Monochloramine solution was prepared as previously described by Chen, Griebe and Characklis (1993) and hydrogen peroxide solutions were prepared by diluting Hach unstabilized hydrogen peroxide (30%) (ACS grade cat # 144-11). Both monochloramine and hydrogen peroxide were titrated prior to each experiment to determine their concentrations. Monochloramine was titrated using a Hach amperometric titrator (Hach, Loveland, CO, USA). To determine hydrogen peroxide concentration, the method of Klothoff and Sandell (1952) was modified by titrating to a colourless endpoint using standardized 0.1 eq l⁻¹ sodium thiosulphate (Na₂S₂O₃).

Planktonic disinfection assay

An overnight culture of *Ps. aeruginosa* was sub-cultured into fresh R2A to a final cell concentration between 10⁷ and 10⁸ c.f.u. ml⁻¹ and incubated for 4 h. The exponentially growing culture of *Ps. aeruginosa* was homogenized with a tissue homogenizer for one minute on ice. The cells were then diluted to 10⁶ c.f.u. ml⁻¹ in phosphate-buffered water (pH 7) (PBW) (American Public Health Association *et al.* 1995). Monochloramine stock solution was added to the PBW to attain a final concentration of 2 mg l⁻¹. Cells were sampled at timed intervals over a 10-min period from the start of the experiment. Cells were removed from the monochloramine and diluted into PBS containing Na₂S₂O₃ (1 mmol l⁻¹ final concentration) to neutralize the monochloramine. Total remaining chlorine in the reaction vessel was determined using the Hach amperometric titrator. Disinfection experiments were also performed with 600 mg l⁻¹ hydrogen peroxide. Cells were sampled every 10 min for 1 h. The reaction was stopped by using 4.12 mmol l⁻¹, final concentration, of Na₂S₂O₃ in PBS. Final hydrogen peroxide concentrations were determined by titration with Na₂S₂O₃ as described above.

As controls, planktonic cells were sub-cultured in PBW (10⁶ c.f.u. ml⁻¹) without the addition of an antimicrobial agent and sampled at the start and end of the disinfection assay. For all disinfection experiments, cells were serially diluted in PBS and enumerated by the drop plate method, as described above.

Biofilm disinfection assays

Alginate beads with attached cells were removed from the R2A, rinsed three times with sterile medium, and suspended

in 1000 ml of PBW. While mixing beads on a magnetic stir plate, monochloramine was added to reach a final concentration of 2 mg l⁻¹. Between 30 and 50 beads were removed with a wide-mouth 5-ml pipette tip at timed intervals over a 10-min period and neutralized with Na₂S₂O₃ (1 mmol l⁻¹ final concentration). The supernatant was decanted from the bead sample and 3 ml of citrate buffer (pH 7.5) (8 g of sodium citrate in 1 l of PBS) added to facilitate bead dissolution. Beads were blended on ice using a tissue homogenizer for 1 min and refrigerated for up to 2 h. Total remaining chlorine in the reaction vessel was titrated as described above. Biofilm disinfection studies were also performed with hydrogen peroxide. This disinfectant was added to a final concentration of 600 mg l⁻¹, and beads were sampled every 10 min for 1 h. Supernatant was decanted immediately after bead removal from the reactor and beads were neutralized with 3 ml of 6.32 μmol l⁻¹ Na₂S₂O₃ in citrate buffer. Beads were homogenized on ice and stored at 4 °C for up to 2 h.

As controls, attached cells were diluted in PBW as described above, but without the addition of an antimicrobial agent. These were sampled at the beginning and the end of the assay. In addition, disinfection experiments were repeated with sterile beads to ensure that beads did not significantly degrade monochloramine or hydrogen peroxide. Homogenized beads were diluted in PBS and enumerated.

Biofilm formation and disinfection on glass slides

Ps. aeruginosa biofilms were grown on glass slides in a biofilm apparatus described by Cargill *et al.* (1992). Biofilms were grown for 24 h in conditions similar to the bead biofilms, in that spent medium was removed every 4–8 h to maintain nutrient replete growth conditions. Four hours before disinfection assays, slides were placed into fresh R2A. Disinfection of the glass slides occurred in a clean sterile jar to eliminate the oxidant demand associated with excessive organic matter. Sample slides were removed at timed intervals and placed into PBW containing sodium Na₂S₂O₃ to neutralize the biocide. Slides were then scraped into PBS and cells were homogenized and enumerated on R2A plates.

Disinfection rate coefficient values

The model used to interpret disinfection rates was

$$dX/dt = -k_b CX$$

where the change in areal cell density with time is related to the disinfection coefficient, k_b , biocide concentration, C , and viable cell density, X . Raw disinfection rates were determined by using the least squares method to calculate a regression line through the data consisting of the natural logarithm of c.f.u. cm⁻² vs time (units of min⁻¹). A slope with its standard

error was calculated, where the slope was the disinfection rate for that experiment. The disinfection rate coefficient was determined by dividing raw disinfection rates by the average biocide concentration (units mg l^{-1}) to normalize against variations in disinfectant concentration. This coefficient has units of $\text{l mg}^{-1} \text{min}^{-1}$.

Statistical analysis

All statistical analyses were performed using Minitab Release 11.12 (Minitab Inc., State College, PA, USA). *P*-values were calculated using the null hypothesis that the two sets of disinfection rate coefficients were the same. A *P*-value less than 0.05 indicates that the two sets are statistically different.

Observed penetration of hydrogen peroxide into alginate gel beads

Mass transport of hydrogen peroxide into gel beads was visualized experimentally. Beads with 24-h-old biofilms were incubated with hydrogen peroxide colour indicators used in hydrogen peroxide titration, as described above, for 15 min. Colour indicator solution was removed and 600 mg l^{-1} hydrogen peroxide was added to the beads. Images were digitally captured at timed intervals to show the diffusion of hydrogen peroxide into the beads. A COHU[®] camera (Cohu Inc., San Diego, CA, USA; model no: 2222-1040/0000) and Flashpoint[®] frame grabber (Integral Technologies Inc., Indianapolis, IN, USA) connected to a computer with ImagePro 3.0 (Media Cybernetics, Silver Spring, MD, USA) were used.

Mass transport analysis

To address how effectively solutes (biocides or nutrients) added to the bulk solution were transported to the bead surface, mass transport limitation was theoretically analysed. This was done by calculating the time required to attain 90% of the bulk fluid concentration at the bead surface or, in the case of a sustained neutralization or utilization reaction by the microbial cells, the concentration of biocide at the bead surface that was attained at steady state.

The unsteady diffusion of a non-reacting solute from the bulk fluid into the permeable gel bead was addressed first. The diffusion equation in spherical co-ordinates was solved subject to a matching flux boundary condition and an initial condition imposing zero concentration throughout the bead. The solution to this problem was given by Crank (1956).

Parameter values required for these calculations include the bead dimension, effective diffusion coefficients, and an external mass transfer coefficient. A bead radius of 0.1 cm was used. Diffusion coefficients in gel beads were taken as 90% (Westrin and Axelsson 1991) of the value in water at

25 °C as estimated from the Wilke-Chang correlation (Perry and Chilton 1973). The effective diffusion coefficient of monochloramine in the gel bead calculated in this way was $1.7 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$ and for hydrogen peroxide it was $1.3 \times 10^{-5} \text{ cm}^2 \text{ s}^{-1}$. The mass transfer coefficient describing transport between the bulk fluid and the bead surface was estimated by Xu *et al.* (1996), using an established correlation, to be $5 \times 10^{-3} \text{ cm s}^{-1}$.

Bacteria neutralize monochloramine and hydrogen peroxide (Brown *et al.* 1995; Sanderson and Stewart 1997) and utilize nutrient substrates. Bacteria attached to a substratum can reduce the local concentration of these molecules at the bead surface below the bulk fluid concentration. This problem can be mathematically analysed by equating the flux due to reaction with the flux determined by the external mass transfer coefficient for the system. From Sanderson and Stewart (1997), the order of magnitude of *Ps. aeruginosa* capacity for reaction with monochloramine was $5 \times 10^{-14} \text{ mg s}^{-1} \text{ cell}^{-1}$. Taking the highest cell density encountered in the present experiments of $4 \times 10^5 \text{ c.f.u. cm}^{-2}$, the maximum flux due to reaction was approximately $2 \times 10^{-8} \text{ mg cm}^{-2} \text{ s}^{-1}$. At steady state this must exactly match the flux supplied by transport from the bulk fluid to the surface and the concentration at the bead surface can be solved. An analogous calculation was performed for hydrogen peroxide using the induced level of catalase activity reported by Brown *et al.* (1995), equivalent to approximately $2 \times 10^{-10} \text{ mg s}^{-1} \text{ cell}^{-1}$, and for oxygen using $1 \times 10^{-13} \text{ mg s}^{-1} \text{ cell}^{-1}$ (Bakke *et al.* 1984).

RESULTS

Disinfection rate coefficients

Planktonic disinfection rate coefficients for monochloramine experiments all fell within a tight cluster of values ranging from 0.40 to $0.571 \text{ mg}^{-1} \text{ min}^{-1}$ (Fig. 1). Disinfection rate coefficients were calculated by dividing the raw disinfection rates (min^{-1}) by the average biocide concentration (mg l^{-1}). This normalized the data against any experiment-to-experiment variation in the biocide concentration. Disinfection rate coefficient values were significantly higher for planktonic cells than they were for 20-h-old and older biofilms, for which disinfection rate coefficient values fell into a range of 0.19–0.231 $\text{mg}^{-1} \text{ min}^{-1}$. *P*-values comparing planktonic disinfection rate coefficient values with 20-h-old and older biofilms were all less than 0.008 (Table 1).

There was no correlation between initial cell density of the biofilm and disinfection rate coefficient (Fig. 2). The areal cell density of the gel bead-attached biofilms prior to treatment was relatively low averaging $3.5 \pm 0.58 \log \text{ c.f.u. cm}^{-2}$. The *P*-value for the regression line fit to the data in Fig. 2 was 0.84, indicating that there was no significant dependence

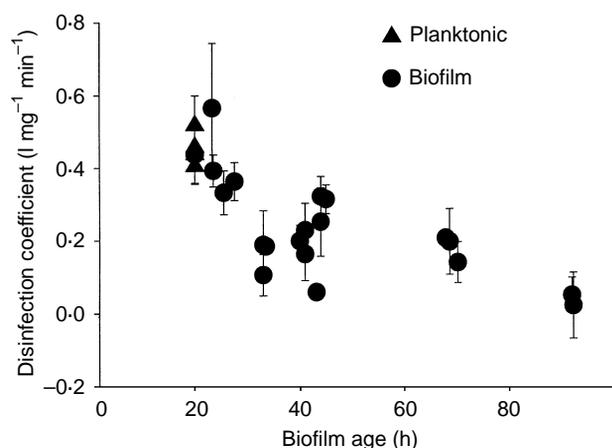


Fig. 1 Disinfection rate coefficients of planktonic and biofilm *Ps. aeruginosa* cells treated with 2 mg l^{-1} of monochloramine. Error bars represent standard errors

Table 1 *P*-values for comparison of planktonic and biofilm susceptibility

	Biofilm age (h)			
	20	24	48	72
	$n = 3^*$	$n = 3^*, \dagger$	$n = 3^*, 2\dagger$	$n = 2^*$
<i>Biocide</i>				
MCA	0.0002	0.0018	0.0003	0.0004
H ₂ O ₂		0.015	0.019	

n is the number of replicates for biocide experiments.

* Replicate number for monochloramine experiments.

† Replicate number for hydrogen peroxide experiments.

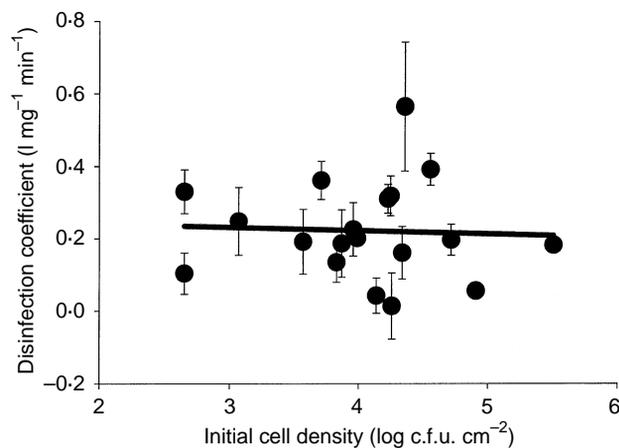


Fig. 2 Influence of initial cell density on disinfection efficacy. Error bars represent standard errors

of biofilm susceptibility to monochloramine on the initial cell areal density.

Disinfection with hydrogen peroxide (Fig. 3) yielded results qualitatively similar to the monochloramine experiments. Planktonic cell disinfection rate coefficients ranged from 2.3×10^{-4} to $4.0 \times 10^{-4} \text{ l mg}^{-1} \text{ min}^{-1}$ and biofilm disinfection rate coefficients ranged from 8.8×10^{-5} to $9.7 \times 10^{-5} \text{ l mg}^{-1} \text{ min}^{-1}$. The planktonic disinfection rate coefficient values were significantly greater than the biofilm disinfection rate coefficient values with *P*-values less than 0.05 for these comparisons (Table 1). As with the monochloramine data, there was no correlation between initial density of biofilm cells and disinfection rate coefficient values when using hydrogen peroxide ($P = 0.43$).

Additional disinfection studies with monochloramine and hydrogen peroxide were performed using biofilms grown on glass slides instead of on alginate gel beads. Biofilm growth on glass slides resulted in thicker biofilms than biofilms grown on alginate beads. The initial areal cell density of glass slide biofilms averaged of $7.6 \pm 0.27 \text{ log c.f.u. cm}^{-2}$. These biofilms were also clearly less susceptible to both disinfectants when compared with planktonic cells (Fig. 4). Disinfection rate coefficient values for 24-h biofilms ranged from 5.5×10^{-2} to $0.201 \text{ mg}^{-1} \text{ min}^{-1}$ for monochloramine and from 3.4×10^{-5} to $8.0 \times 10^{-5} \text{ l mg}^{-1} \text{ min}^{-1}$ for hydrogen peroxide. These biofilm disinfection rate coefficient values were statistically significantly lower than values for planktonic cells ($P < 0.01$).

Cell attachment and distribution on alginate beads

GFP expressing *Ps. aeruginosa* clearly showed the distribution of attached cells on alginate beads (Fig. 5). Beads with cells attached for 24 h were sparsely populated. Much of the bead

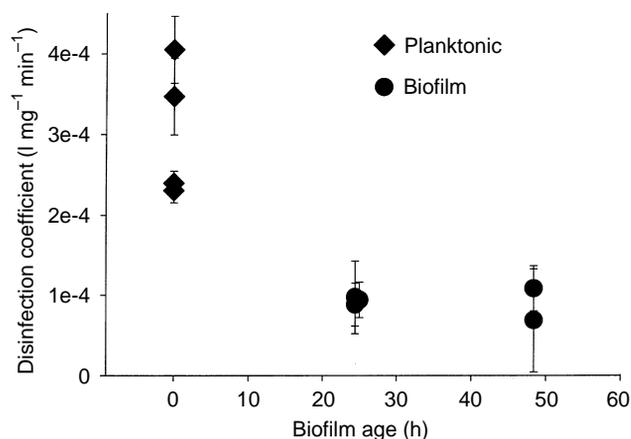


Fig. 3 Disinfection rate coefficients of planktonic and biofilm *Ps. aeruginosa* cells treated with 600 mg l^{-1} of hydrogen peroxide. Error bars represent standard errors

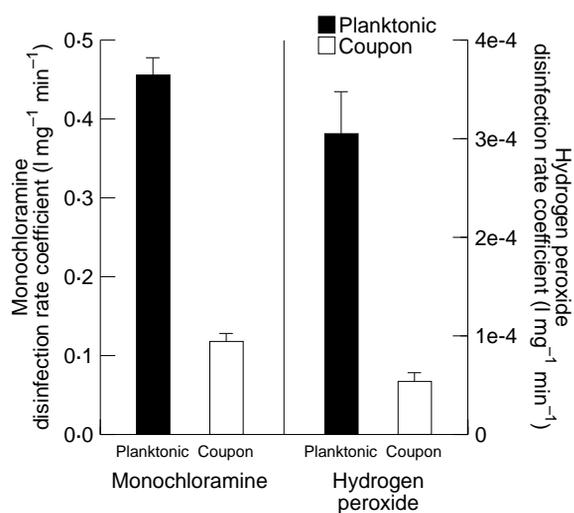


Fig. 4 Mean disinfection rate coefficients obtained from biofilms grown on glass slides exposed to either $2\ mg\ l^{-1}$ of monochloramine or $600\ mg\ l^{-1}$ of hydrogen peroxide. Error bars represent standard error of the means. Each coupon experiment was repeated five times to calculate the mean disinfection rate coefficient

area observed contain few or no cells. The cells that were observed appeared to be attached as single cells or in small microcolonies. Colonies were uncommon. Although, 48-h-old biofilms were slightly more populated, the overall trend of sparsely distributed cells was the same as that observed for the 24-h-old biofilms.

Hydrogen peroxide penetration in alginate gel beads

Hydrogen peroxide was shown to penetrate cell-covered gel beads rapidly by a direct visualization technique. Gel beads incubated in hydrogen peroxide titration reagents turned a characteristic blue-black colour upon the addition of hydrogen peroxide. The rapid appearance of this colour at the surface of the gel beads and in progressively deeper layers of the gel bead could be seen by examination with low power magnification (Fig. 6). Full penetration was observed within 2.5 min after the addition of hydrogen peroxide (Fig. 6c). When the neutralizing agent sodium thiosulphate was added to the beads, the bead colour turned back to a translucent white from the outside to the inside as the neutralizer diffused through the bead. Within 5 min after the sodium thiosulphate addition, hydrogen peroxide in the gel bead had been completely neutralized, as indicated by a uniform white colour throughout the bead (Fig. 6f).

Theoretical analysis of mass transport

Penetration of biocides to the bead surface was not significantly hindered by mass transport according to theoretical



Fig. 5 Use of GFP to detect *Ps. aeruginosa* biofilms on alginate gel beads. Epifluorescence of 24-h-old biofilms showed that cell distribution consisted mostly of single cells and small microcolonies. Bar represents $5\ \mu m$

calculations. An unsteady diffusion model predicted that after 33 s the monochloramine concentration at the bead surface would be 95% of the bulk fluid concentration. This was relatively rapid when compared with the duration of the disinfection experiments of 6–10 min. For hydrogen peroxide the calculated time to reach 95% of the bulk fluid concentration at the bead surface was 44 s. Again, this represented only a brief delay compared with the hour-long treatment period. When a biocide-neutralizing reaction was taken into account, the steady-state bead surface concentration was predicted to be at least 99.8% of the bulk fluid concentration for monochloramine and at least 96% of the bulk fluid concentration for hydrogen peroxide. In summary, theoretical analysis of biocide transport suggests that transport limitation can explain no more than a few percent of the observed reduction in susceptibility of biofilm bacteria.

The calculated oxygen concentration at the bead surface was greater than 99% of the bulk fluid oxygen concentration.

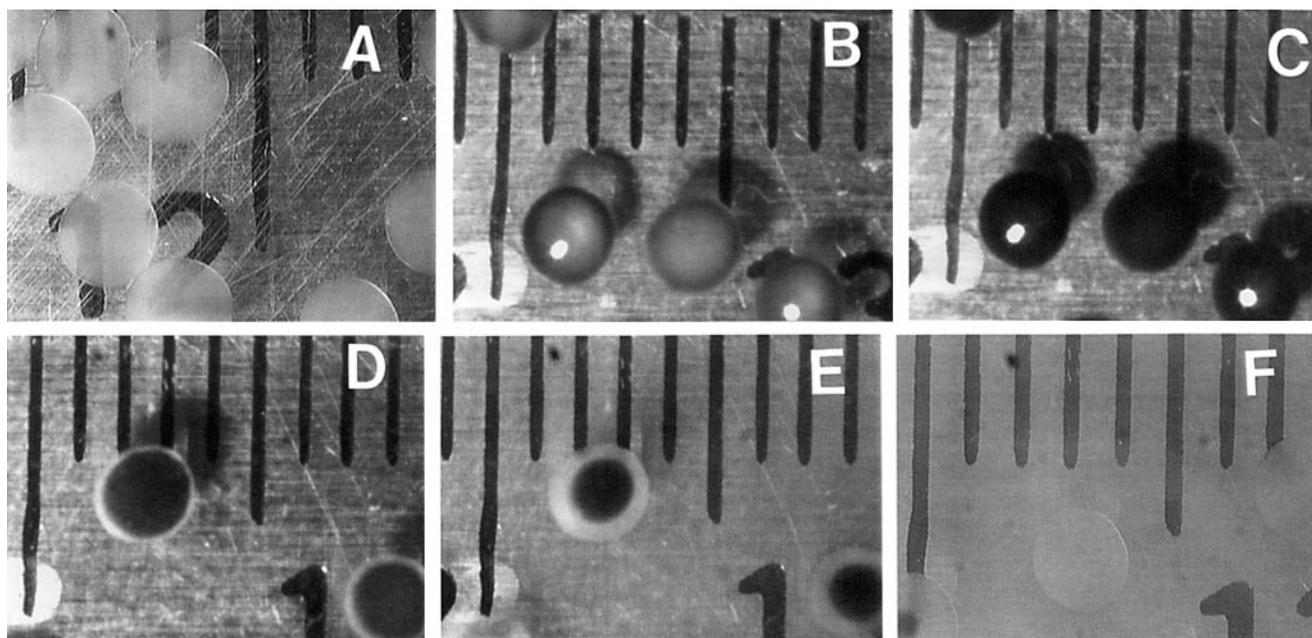


Fig. 6 Diffusion of hydrogen peroxide into 24-h gel bead biofilms. Beads were incubated in hydrogen peroxide colour indicators for 15 min. 600 mg l^{-1} of hydrogen peroxide was then added. Photographs were taken at 0 s (A), 10 s (B) and 150 s (C). As hydrogen peroxide diffused through the bead, the bead turned black. Sodium thiosulphate, 0.1 eq l^{-1} was added to the beads to show the diffusion of the neutralizer into the gel beads. Photographs shown were taken at 10 s (D), 120 s (E) and 290 s (F) after the addition. As sodium thiosulphate diffused through the bead, the hydrogen peroxide was neutralized and bead turned back to white

Local anaerobic conditions in the biofilm can therefore not be anticipated.

DISCUSSION

Reduced susceptibility of biofilm cells

Ps. aeruginosa cells attached to alginate gel beads were significantly less susceptible to disinfection by monochloramine and hydrogen peroxide than were planktonic cells of the same micro-organism. The observed resistance increased with increasing biofilm age, at least for monochloramine. Reduced susceptibility could not be detected with statistical significance for biofilm cells less than 13-h-old but was clearly established by the time cells had been attached for 24 h.

Reduced susceptibility is not a transport artifact

Several pieces of evidence support the interpretation that reduced susceptibility of biofilm was not simply due to inadequate delivery of the antimicrobial agent to the local environment of the attached cells. Solute penetration into a biofilm depends critically on the biofilm thickness. If antimicrobial penetration was a governing factor, one would expect to see an inverse correlation between biofilm susceptibility and biofilm areal cell density. No such correlation was observed

(Fig. 2). Direct microscopic observation of cells attached to the gel beads confirms that cells are distributed in a sparse, thin layer (Fig. 5). Visualization of the rapid delivery of hydrogen peroxide to the surface, and subsequently to the interior, of the alginate gel beads (Fig. 6) provides a direct experimental demonstration of effective biocide penetration. Finally, theoretical analysis of unsteady diffusion and diffusion-reaction interactions argue against any significant delay or barrier to antimicrobial delivery.

Similarly, transport limitation of a nutrient leading to a stationary phase population cannot be invoked to explain reduced susceptibility of biofilm in this system. In the case of an obligate aerobe such as *Ps. aeruginosa*, the low solubility of oxygen and the known stoichiometry of aerobic growth ensure that oxygen will be the first substrate to be depleted in the biofilm provided that the carbonaceous substrate concentration exceeds approximately 20 mg l^{-1} . An order of magnitude calculation of the balance of reaction and diffusion for oxygen in this system showed that the reduction in oxygen concentration at the bead surface would be less than 1%. In summary, our alginate gel bead biofilms are simply too thin and the mass transfer rates between the suspended gel beads and the bulk fluid too fast to give rise to significant transport artifacts of any kind.

Biofilms grown on glass slides were orders of magnitude thicker than alginate gel bead biofilms. Glass slide-grown

biofilms also exhibited reduced susceptibility to disinfection by both monochloramine and hydrogen peroxide. These biofilms were probably protected by multiple resistance mechanisms, including transport limitation of biocide penetration into the biofilm. It is not possible to perform rigorous transport calculations in this case because the external mass transfer coefficient for this system is not easily estimated. This inability to quantify the transport process is a serious limitation that the glass slide biofilm system shares with most other laboratory model systems for measuring antimicrobial efficacy.

Resistant biofilm phenotype hypothesized

Previous studies have shown that when cells attach to a surface they undergo physiological and metabolic changes (Davies and McFeters 1988; McFeters *et al.* 1990; Griffith and Fletcher 1991; Davies, Chakrabarty and Geesey 1993; Vandevivere and Kirchman 1993; Ascon-Cabrera, Ascon-Reyes and Lebeault 1995; Wentland *et al.* 1996). The work reported in this paper shows that reduced susceptibility of attached *Ps. aeruginosa* cells to two oxidative biocides should also be recognized as an inherent phenotypic change and not just a transport artifact. Other studies of thin biofilm susceptibility to various antimicrobial agents support this interpretation (e.g. Das *et al.* 1998), though they lack characterization of the transport processes. If phenotypic changes indeed take place upon cell attachment to a surface, it is natural to wonder to what extent these changes are encoded in the genome. Recent reports are now providing the first glimpse of the genetic basis for biofilm formation (Davies *et al.* 1993; Davies and Geesey 1995; Davies *et al.* 1998; McKenney *et al.* 1998; O'Toole and Kolter 1998; Pratt and Kolter 1998; Stickler *et al.* 1998; Vidal *et al.* 1998). We hypothesize that new genes are expressed when bacteria attach to a surface and begin to form a biofilm and that some of the resulting gene products reduce the susceptibility of the cell to antimicrobial agents including oxidative biocides such as monochloramine and hydrogen peroxide.

In practical terms this means that the problem of biofilm control by antimicrobial agents is indeed complex. In addition to overcoming antimicrobial penetration failure and surmounting regions of starved or nutrient-limited cells, we must also cope with an inherently resistant biofilm phenotype. However, the possibility that the biofilm phenotype is genetically determined may foretell of a multitude of new chemotherapeutic targets for controlling biofilm fouling and infections.

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